

ORIENTAL JOURNAL OF CHEMISTRY

An International Open Access, Peer Reviewed Research Journal

www.orientjchem.org

ISSN: 0970-020 X CODEN: OJCHEG 2021, Vol. 37, No.(6): Pg. 1452-1457

A Facile Synthesis and Characterization of new Nitrogen, Phosphorus, Potassium (N-P-K) Fertilizer Fortified with Tri-micronutrient Matrix and its application for Optimal Plant Augmentation

N. V. S. VENUGOPAL^{1*} and G. N. V. MOHANA RAO²

^{1,2}Department of Chemistry, Institute of Science, GITAM University Visakhapatnam-530045, Andhra Pradesh, India.
*Corresponding author E-mail: vnutulap@gitam.edu

http://dx.doi.org/10.13005/ojc/370626

(Received: October 18, 2021; Accepted: December 14, 2021)

ABSTRACT

At present in agricultural techniques engaged the optimal nutrient supply is very imperative factor for elevated crop yield and all essential plant macro and micronutrients. Micronutrients play vital role in photosynthesis. For balanced crop nutrition micronutrient support is essential. Micronutrient fertilizer required in small quantities by plants and occupy vital role in physiology of plant kingdom. Proven fact is decisive plant functions are over involved if enough micronutrients are engaged. In this paper the author developed a new N-P-K fertilizer fortified with tri-micronutrient matrix and its application for optimal Plant growth. The author developed and analyzed hundred percent water soluble active fertilizer(15-15-15) with three important micronutrients (-3.4 0.5% Zn, 0.5% Fe, 0.5% Mg). Pot experiments were conducted with and without proposed fertilizer on *Solanum lycopersicum* (tomato) seeds. The proposed enhanced efficiency fertilizer with tri-micronutrient matrix showed better plant growth as compared to normal untreated fertilizer in low nutrient soil field.

Keywords: Nitrogen, Phosphorus, Potassium fertilizer, Micronutrient matrix, Pot experiments, Plant growth, *Solanum lycopersicum* seeds.

INTRODUCTION

Food security is one of the world's big challenges with huge demand forecast. Every year a series of agricultural products, drink and food products for human consumption and animal feed are formed. Climatic conditions and total geography have big impact on agricultural use of available land. Various materials added to the soil in order to supply different chemical elements for better soil fertility. Fertilizers are needed for receiving increase in crop growth and yield. Macronutrients such as Nitrogen, Phosphorous and Potassium were recognized for the supply of important nutrients to all crops. At present more chemical fertilizers used by farmers to produce higher yield and augment reasonable efficiency. The use efficiency of N-P-K is about 30-60%; 30-50% and 10-20%, respectively and these reflect low macronutrients use efficiency for crops. The fertilizers not in use by crops instead

This is an <a>Open Access article licensed under a Creative Commons license: Attribution 4.0 International (CC- BY). Published by Oriental Scientific Publishing Company © 2018



entering in to the environment is the pressing concern to the world.

Visual symptoms on crops always reveal micronutrients deficiency and it can be tested from plant tissues and soil. Micronutrients like Iron, Zinc and Magnesium have important role for plant growth. Iron is very important for biological process and cell growth in plants. The main component of enzymes indispensable for chlorophyll synthesis, photosynthesis etc is Iron. The role of Zinc is vital for plant hormone balance and auxin activity. Magnesium is the central core of the chlorophyll molecule in plant tissue. Shortage of chlorophyll results low plant growth if Magnesium is deficient. Magnesium deficiency leads to boost contribution of production and also diminishes cost-effective efficiency and augments environmental pollution.¹⁻²

The recent scenario was in the preparation and application of nanofertilizers. Nanofertilizers help increase the effectiveness of fertilizer, enhance yield and excellence of crops, diminish detrimental effects of chemical fertilizers on environment and develop a green and sustainable agriculture³⁻¹². By coating urea into nanofilm (nanourea) has been triumphant using it for Canola¹³. To enhance yield and reduce nitrate leaching, Nano Nitrogen fertilizer synthesized by coating Urea with Sulfur and nano Nitrogen chelate. Nano Nitrogen was effectively applied for potato. The current research in progress by many researchers in the world was to develop Complex NPK nanofertilizer. Wu and Liu investigated the encapsulation of NPK fertilizer into chitosan and coating the outer by poly (acrylic acid-co-acrylamide). Complex NPK nanofertilizer was prepared by trapping the fertilzer in polyacrylic hydrogel and later the study of the NPK slow release control of the nanofertilizer¹⁴. For loading nutrients to various crops materials like Chitosan and its nanoparticles etc were functional as a useful matrix. Complex NPK fertilizer was encumbered into chitosan nanoparticles, and chitosan nanoparticles with polymethacrylic acid (MMA)¹⁵⁻¹⁹ Therefore from the past forty years chemical fertilizers have been crucial to the growth of world agricultural production.

MATERIALS AND METHODS

The various raw materials used in the current study were given in Table 1. All other chemicals used were of analytical grade.

Table1: Raw materials

S.no	Materials	Molecular formula	Grade
1	Ammonium sulphate	(NH ₄) ₂ SO ₄	Fertilizer grade
2	Mono ammonium phosphate	NH₄H₂PO₄	Fertilizer grade
3	Urea	NH,CO NH,	Fertilizer grade
4	Potassium nitrate	KNO ₃	Fertilizer grade
5	Chelated Zinc	Zn EDTA	Fertilizer grade
6	Ferrous sulphate	FeSO ₄ .7H ₂ O	Fertilizer grade
7	Magnesium sulphate	$MgSO_4.6H_2O$	Fertilizer grade

The above-mentioned raw materials of fertilizers grade were analyzed by using existing methods defined in Fertilizer control order (FCO) procedures to know the nutrient contents accordingly final product formulation prepared. The methods of analysis used for the analysis of raw materials were given in Table 2.

Table 2: Methods of analysis used for the analysis of raw materials

S.no	Parameter	Procedure No
1	Ammoniacal nitrogen	FCO 1985 as amended upto
		February 2019, Schedule
		Part-B-3(vii)
2	Nitrate nitrogen	FCO 1985 as amended upto
		February 2019, Schedule
_		Part-B-3(ix)
3	Urea nitrogen	FCO 1985 as amended upto
		February 2019, Schedule
		Part-B-3(xi)
4	water soluble phosphate	FCO 1985 as amended upto
		February 2019, Schedule
-	Matar aduble pataosium	Part-B-4(III)
э	water soluble potassium	FCO 1985 as amended upto
		Port P 5(ii)
6	Sulphate sulfur	FCO 1985 as amended unto
0	Sulphate Sullui	February 2019, Schedule
		Part-B-24(Δ)
7	Zinc	FCO 1985 as amended upto
		February 2019, Schedule
		Part-B-15
8	Iron	FCO 1985 as amended upto
		February 2019, Schedule
		Part-B-13

Instrumentation

Inductively coupled plasma optical emission spectrometry(ICP-OES), Perkin-Elmer (Optima7000DV) was preferred for elemental analysis. For macronutrients, Auto analyzer (Skalar) was used. Analytical balance, Top load balance (Sartorius), Hot air oven (Tempo), etc were used in the study.

METHODS

All the standard solutions were standardized as per procedure before proceeding to analysis. Performed the calibration to weighing balances and hot air oven at the time of analysis. Blank estimation was performed for all the parameters on the reagents and accordingly correction given to the samples analysis. Duplicate determination was performed for each parameter in the analysis and average was taken as final result.

Preparation of N-P-K fertilizer fortified with tri-micronutrient grade

Each substance-Mono Ammonium Phosphate, Ammonium Sulphate, Urea, Potassium Nitrate, Zinc EDTA, Ferrous Sulphate hepta hydrate and Magnesium Sulphate hexa hydrate have taken 100 g grounded individually with help of Mortar and pestle and kept for drying in a vacuum desiccators for 24 hours. after 24 h the substances are properly kept in self sealed covers to avoid moisture absorption. Three replicate mixtures were prepared separately and each replicate consists of 25.0 g of Mono Ammonium Phosphate, 19.29 g of Ammonium Sulphate, 10 g of Urea, 33.71 g of Potassium Nitrate, 4.17 g of Zinc EDTA, 2.63 g of Ferrous sulphate hepta hydrate and 5.21 g of magnesium sulphate hexa hydrate. All the mixtures are finely grounded and mixed for 20 min in a mixer of Preeti make, model no MG 198 (1300 watts) with 4000rpm by using small jar of capacity 700 mL. All the three replicate mixtures are dried in a vacuum desiccators for 24 h to remove surface moisture absorbed during the preparation process.

RESULTS AND DISCUSSION

Initially all the raw materials were analyzed and the results are given in Table 3.

Table 3: Raw material analysis

Name of the raw material	ercentage obtained
Ammonium Sulphate (Ammonical Nitrogen-AN Sulphate sulphur as S Mono Ammonium Phosphate: Ammonical Nitrogen-N Water soluble Phosphate as P_2O_5 Urea:Urea nitrogen as N Potassium Nitrate:water soluble as K_2O Nitrate Nitrogen as N Chelated Zinc-EDTA:as Zn	20.2 23.2 11.5 60.2 46.1 44.5 12.5 12.1
Magnesium Sulphate hexa Hydrate:as Mg	9.6

A new formulated 100% water soluble fertilizer with three important micronutrients (15-15-15-3.4 0.5% Zn, 0.5% Fe, 0.5% Mg) was developed and analyzed. The N-P-K fertilizer fortified with tri-micronutrient product was shown in Figure 1.



Fig. 1. N-P-K fertilizer fortified with tri-micronutrient Analysis

- Total Nitrogen and Nitrate nitrogen are analyzed by macro Kjedhal distillation method as per FCO 1985 procedures.
- ii) Water soluble Potassium was analyzed by STPB (sodium tetra phenyl boran) method prescribed in FCO1985. For this taken 2.5 g sample and dissolved in Ammonium oxalate, and from that required aliquot taken and added 2 mL 20% NaOH, 5 mL HCHO and 25 mL STPB reagent and made upto 100 mL mark with distilled water. Filtered through no. 42 Whatman filter paper. 50 mL filtrate was taken in 150 mL conical flask titrated against Cetyl trimethyl ammonium bromide. Blank and standard KH_2PO_4 determination was also done and accordingly correction given.

Sulphate sulphur was determined in the samples by digesting the samples with Concentrated HCl, filtered and precipitated as Barium sulphate with Barium chloride solution, dried at 250°C, weighed and calculated as Sulphur.

iv) Ammoniacal nitrogen (AN), Urea Nitrogen (UN), Water Soluble phosphate(P₂O₅) were analyzed by segmented flow analyzer also called Auto analyzer.

Sample preparation: Three replicates were analyzed for physical parameters, nutrients and heavy metals present in the new proposed fertilizer grade. Known quantity of the sample dissolved in distilled water,

vi)

filtered, and analyzed on segmented flow analyzer (SKALAR SAN++ system).

Principles involved in the analysis on the auto analyzer

AN determination

The reaction involved is Berthlot reaction in which Ammonia reacts with sodium hypo chlorite in presence of buffer and forms monochloramine which reacts with salicylate to form 5-aminosalicylate. A green colored complex is formed after oxidation and oxidative coupling which is then measured at 660nm.

WSP₂O₅ determination

The sample is mixed with ammonium heptamolybdate solution and forms molybdo phosphoric aicd which reacts with metavanadate to form stable yellow coloured vanadomolybdo phosphoric acid, measured at 420nm.

UN determination

The sample is mixed with diacetyl monoxime at 90°C, forms coloured complex, which is further intensified by the addition of thiosemicarbazide and an acid reagent is used to increase the rate of colour formation, measured at 520nm.

A combined stock standard solution of 10,000ppm P_2O_5 , 4000ppm AN, 4660ppm UN prepared by taking appropriate weights of AR grade NH₄Cl, KH₂PO₄, and Urea. From this stock solution, three working standards (S1, S2, S3) prepared, in which S1 contains 500ppm P_2O_5 , 233ppm UN, 200ppm AN; S2 contains 1000ppm P_2O_5 , 466ppm UN, 400ppm AN; and S3 contains 1500ppm P_2O_5 , 699ppm UN, 600ppm AN. Analyzer was calibrated with above 3 working standards in first order reaction mode. Correlation coefficient for all three parameters is above 0.999 and RSD is <0.01% (shown in Figure 2, 3 & 4).









The samples were strongly digested with HCI and HNO₃ and then diluted with Millipore ultra-pure distilled water (conductivity<0.1microsiemens/cm²). Calibration was done with three multi element working standards of 10,50,100ppm which were prepared from stock 1000ppm ICP multi element standard solution IV (Merck, certipur). Same acid back ground maintained for working standards used for calibration. The wave lengths selected for analysis of Fe is 238.204nm, Zn is 213.857nm and Mg is 278.297nm. Correlation coefficient was above 0.999 and RSD was less than 1.0%.

Heavy metals-As, Pb & Cd were analyzed on ICP OES (OPTIMA 7000DV).

The samples were strongly digested with HCl and HNO_3 of trace metal grade, then diluted with Millipore ultra-pure distilled water. Calibration was done with three multi element working standards of 0.25,0.5 and 1.0ppm which were prepared from stock 100ppm ICP multi element standard solution X (Merck, certipur).). Same acid back ground maintained for working standards used for calibration. The wave lengths selected for analysis of As is 188.979nm, Pb is 220.350nm and Cd is 226.50nm. Correlation coefficient was observed above 0.999 and RSD was less than 5.0%.

Test parameter	Result		
·	Trail-1	Trail-2	Trail-3
Total Nitrogen per cent by weight, minimum	15.1	15	15.2
Nitrate nitrogen per cent by weight, minimum	4.1	3.9	4.0
Ammoniacal nitrogen per cent by weight, minimum	6.32	6.4	6.5
Urea Nitrogen percent by weight, minimum	4.65	4.7	4.7
Water Soluble phosphate (as P ₂ O ₅) per cent by weight, minimum	15.1	15.2	15.0
Water soluble potash (as K,O) per cent by weight, minimum	15	15.1	15.15
Sulphate sulfur (as S) percent by weight, minimum	3.5	3.5	3.53
Zinc as Zn per cent by weight, minimum	0.53	0.5	0.52
Iron as Fe per cent by weight, minimum	0.51	0.49	0.51
Magnesium as Mg per cent by weight, minimum	0.51	0.5	0.50
Moisture per cent by weight, maximum	0.8	0.7	0.75
Heavy metals content			
Arsenic as As ppm by weight, maximum	4.0	3.0	3.4
Lead as Pb ppm by weight, maximum	<1	<1	<1
Cadmium as Cd ppm by weight, maximum	<1	<1	<1

Table 4: Analysis of N-P-K fertilizer fortified with tri-micronutrient

Application of the proposed fertilizer to Solanum lycopersicum seeds

The investigation of Solanum lycopersicum plants under controlled conditions was carried without off-putting effects of various environmental aspects. Pot experiments were conducted with and without the proposed fertilizer. In one pot soil weight of 955 g and 100 mL of water and gently mixed and on the top of the soil 0.03 g of *Solanum lycopersicum* seeds applied uniformly. In another pot, soil weight of 954 g and 100 mL of 1% proposed fertilizer in water are gently mixed to get uniformity in the soil and on the top of the soil 0.03 g of *Solanum lycopersicum* seeds applied uniformly. The growth pattern was shown in Figure 5.



Fig. 5. Growth pattern of Solanum lycopersicum

CONCLUSION

compared to normal low nutrient soil field.

ACKNOWLEDGEMENT

All crops require important macronutrients such as N-P-K. Fertilizers are crucial to different crops, endow with obligatory nutrients for crop growth, and increase crop yield and quality of the crop. N-P-K fertilizer fortified with tri-micronutrient matrix and its application for optimal Plant growth shown best results and more crop yield. The proposed fertilizer showed better plant growth as

The author wish to thank the management of GITAM (Deemed to be University), Visakhapatnam, Andhra Pradesh, India for supporting this work.

Conflict of Interest

The authors declare that there are no conflicts of interest.

REFERENCES

- 1. Sun, B.; Zhang, L.; Yang, L.; Zhang, F.; *Ambio.*, **2012**, *41*, 370.
- 2. Trenkel, M.M, Controlled release and stabilized fertilizer in agriculture (International Fertilizer Industry Association, Paris., **2010**.
- Corredor, E.; Testillano, P.S.; José Coronado, M.; González-Melendi, P.; Fernández-Pacheco, R.; Marquina, C.; Ricardo Ibarra, M.; De la Fuente, J.M.; Rubiales, D.; De Luque, A.P.; Carmen Risueño, M.; 2009, *BMC Plant Biol.*, 2009, *9*, 1.
- 4. Ghormade, V.; Deshpande, M.V.; Paknikar, K.M.; *Biotechno.*, **2011**, *29*, 792.
- 5. Huang, S.; Wang, L.; Liu, L.; *Agron. Sustain. Dev.*, **2015**, *35*, 369–400.
- Khodakovskaya, M.; Dervishi, E.; Mahmood, M.; Xu, Y.; Z. Li, F.; Watanabe, A.S.; ACS Nano., 2009, 3, 322.
- 7. Liu, R.; Lal, R.; *Sci Rep.*, **2014**, *4*, 5686.
- Prasad,T.N.V.K.V.; Sudahka, P.; Sreenivasulu, Y.; Latha, P.;Munaswamy, V.; Raja Reddy, K.; Sreeprasat, T.S.; Panikkanvalappil, S.R.; Thalappil, P.; *J. Plant Nutr.*, **2012**, *35*, 905.
- 9. Servin, A.; Elmer, W.; Mukherjee, A.; Hamdi, H.; White, J.C.; Bindranban, P.; Dimkpa, C.,

J. Nanopart. Res., 2015, 17, 92.

- 10. Tarafdar, J.C.; Raliya, R., Mahawar, H.; *Agric Res.*, **2014**, *3*, 257–262.
- 11. Wu, L.; Liu, M.; Carbohydr. Polym., 2008, 72, 240.
- 12. Zareabyaneth, H.; Bayatvarkeshi, M.; crop Environ. Earth Sci., **2015**, *74*, 3385.
- De Rosa, M.C.; Monreal, C.; Schitzer, M.; Walsh, R.; Sultan, Y.; *Nanotechnol.*, **2010**, *5*, 9.
- 14. Teodorescu, M.; Lungu, A.; Stanescu, P.O.; Neamtu, C.; *Ind. Eng. Chem. Res.*, **2009**, *48*, 6527.
- 15. Corradini, E.; de Moura, M.R.; Mattoso, L.H.C.; *Polym. Letters.*, **2010**, *4*, 509.
- 16. Ha, N.M.C.; Nguyen, T.; Wang, H.; *Res Chem Intermed.*, **2019**, *45*, 51–63.
- Phelan, P.; Moloney, A. P.; McGeough, E. J.; Humphreys, J.; Bertilsson, J.; O'Riordan, E.G.; O'Kiely, P.; *Critical Reviews in Plant Sciences.*, **2015**, *34*, 281-326.
- Garde-Cerdan,T.;Santamaría, P.; Rubio-Breton, P.; Gonzalez-Arenzana, L.; Lopez-Alfaro, I.; Lopez, R.; *Food Sci. Technol.*, 2015, 60, 684-689.
- Fernández-Escobar, R.; Antonaya-Baena, M.F.; Sánchez-Zamora, M.A.; Molina- Soria, C.; Scientia Hortic., 2014, 167, 1-4.