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# Facile Synthesis, Characterization and Antimicrobial Activities of Novel 6-Amino triazolo-thiadiazoles Integrated with Benzofuran and Pyrazole Moieties

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### ABSTRACT

In the present research we have reported simple efficient synthesis technique to afford a novel series of 3-(5-(benzofuran-2-yl)-1-phenyl-1*H*-pyrazol-3-yl)-*N*-aryl-[1,2,4]triazolo[3,4-b][1,3,4] thiadiazol-6-amine (4a-g) derivatives obtained by one pot cyclocondensation reaction of 5-(5-(benzofuran-2-yl)-1-phenyl-1*H*-pyrazol-3-yl)-4-amino-4*H*-1,2,4-triazole-3-thiol (3) with substituted aryl isothiocyanate in DMF and  $K_2CO_3$ , without formation undesirable side products by simple work up procedure. The structures yielded (4a-g) were established by <sup>13</sup>CNMR, IR, <sup>1</sup>HNMR, elemental analysis and mass spectra. Entire synthesised compounds were screened for their *In vitro* biological assay via microorganism *Gram-positive* and *Gram-negative* bacterial strains at different concentrations. The bioassay revealed that some of the compounds have promising antimicrobial activities when compared with standard drug Chloramphenicol.

Keywords: Triazolothiadiazole, Pyrazole, Benzofuran-2-yl, Antimicrobial.

# INTRODUCTION

In recent years, fused heterocycles with three hetero atoms in five membered aromatic structures such as 1,2,4-triazoles and thiadiazoles sizable devotion owing to synthetic and remarkable pharmacological activities. The amino triazolothiadiazole system considered as a cyclic crucial and versatile fused five membered heterocycle ring structure incorporating two nitrogen and one sulphur atom. The numbers of triazoles fused to thiadiazoles exhibit various therapeutically important property probably due to the existence of N-C-S fragment in ring. Literature survey has revealed that the 6-amino thiadiazole structure plays vital role in biologically active compounds consequently represents fascinating moiety for therapeutic chemistry.1,3,4thiadiazoles are vital classes of azoles with significant pharmacological activities such antimicrobial<sup>1-6</sup>, antioxidant<sup>7</sup>, antituberculosis<sup>8</sup>, anticancer<sup>9-11</sup>, analgesic<sup>12</sup>, anti-inflammatory<sup>13</sup>, antiviral<sup>14-15</sup>, antifungal<sup>16-19</sup>, antitumor<sup>20</sup>, urease inhibitor<sup>21</sup>, analgesic and anti-inflammatory<sup>22</sup>, antidepressant<sup>23</sup>, anticonvulsant<sup>24</sup>, antimycotic<sup>25</sup>, diuretic<sup>26</sup>, cytotoxic<sup>27</sup>, corrosion inhibition effect<sup>28</sup>, antiproliferative<sup>29</sup>, anthelmintic<sup>30</sup>. Moreover, nowadays researchers

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desire to yield fused or hybrids of various heteroatom ring to improve the medicinal property.

Prompted by these interpretations and in extension of our of determinations towards the synthesis of novel heterocyclic combinations with potent antimicrobial properties, in the present research we planned to frame a molecule and to synthesize and characterize a new series of condensed systems which combines two bio labile rings give a condensed and planar system of triazolothiadiazole with an anticipation to obtain compounds with better enhanced pharmacological activities and further thought of carrying out the antimicrobial studies of these innovative synthesized compounds against some bacterial strains.

## MATERIAL AND METHODS

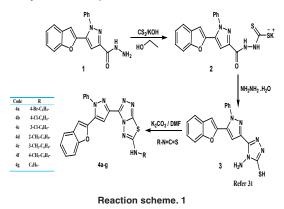
E. Merck TLC aluminium sheet silica gel was used for monitoring reactions; iodine and UV light chamber were utilized for visualizing of spots. Melting points obtained in open capillary tube. Shimadzu IR Spectrophotometer is used to record IR on a (KBr, v max in cm<sup>-1</sup>). ESI mass spectra were noted on Waters Micromass Q–TOF Micro, Mass Spectrophotometer. <sup>1</sup>H NMR and <sup>13</sup>CNMR spectra are logged in Bruker AM instrument having 400 MHz and values reported in (ppm) by (CDCI<sub>3</sub> and DMSO*d6*) in respect to tetramethylsilane (TMS). On Thermo Scientific (Flash-2000) element (CHN) analysis done using all the acquired products screened for their antimicrobial activities.

#### General protocol to yield 3<sup>31</sup>

Suspension of potassium salt (2) (20 mmol), 95% hydrazine hydrate (40 mmol) in  $H_2O$  (2 mL) was taken in R.B flask and heated on condenser with occasional shaking for 1 hours. The colour of the content transformed to greenish with the evolution of  $H_2S$  gas. The resultant solution was added to ice cold water. The solid 3 was separated out by acidification with conc. HCI, filtered, and recrystallization was carried out by using ethanol.

# Synthesis of 6-amino-1,2,4-triazolo[3,4-*b*][1,3,4] thiadiazole (4a-h)

A mixture of compound (3, 3.74 g, 10mmol), and 1-bromo-4-isothiocyanatobenzene (2.14g, 10mmol) in DMF(25 mL), was taken in round bottom flask to that potassium carbonate (1.38 g, 10mmol) was added then the reaction mixture was refluxed for 8 hours. Then reaction content was cooled discharged slowly with stirring into crush ice subsequent product obtained filtered, clean thoroughly with cold water. Correspondingly, 4b-g were synthesised from 3 by extending the same method followed for 4a and their structural identities were proved by chemical transformation reaction, physical data, and elemental analysis and IR spectra.



#### Spectral and Elemental analysis

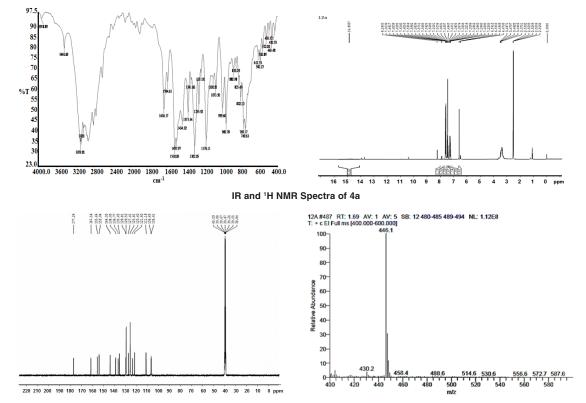
(4a): IR, 3440(N-H),1594,1636(N-H), 3079, 3020(CH), 1497, 1518, 1594 (C=C),1255(C-O-C),1636(C=N), 1255(C-N), 999(N-N), 749(C-S-C) cm<sup>-1</sup>. <sup>1</sup>H NMR  $\delta$  (ppm);14.86(s, 1H, aromatic secondary NH group), 6.59 (s, 1H, at C4 of pyrazole ring), 7.22-7.91(m, 14H, aromatic + Heteroaryl proton).<sup>13</sup>C NMR (DMSO-*d6*):  $\delta$  (ppm)106, 111, 121, 123, 125, 127.41, 129, 135, 136, 138, 144(s, 1C, C<sub>3</sub> of pyrazole ring), 153(s,1C, C<sub>9</sub> of Benzofuran ring), 155, 161(s,1C,C<sub>2</sub> of triazolothiadiazole), 171(s, 1C, C<sub>8</sub> of triazolothiadiazoles).GC-MS (*m/z*): 556 [M+2]<sup>+</sup>. Elemental Analysis for C<sub>26</sub>H<sub>16</sub>BrN<sub>7</sub>OS Calculated, 56.33; H, 2.91; N, 17.68; S, 5.78 Found: C, 56.40; H, 2.98; N, 17.74; S, 5.70.

(4b): IR, 3442(N-H), 1596, 1634(N-H), 3076, 3018(C-H), 1495, 1512, 1596(C=C), 1256(C-O-C), 1634(C=N), 1256(C-N), 1002(N-N), 750(C-S-C) cm<sup>-1</sup>.

(4c): IR, 3441(N-H), 1591, 1638(N-H), 3077, 3023(C-H), 1496, 1516, 1598(C=C), 1259 (C-O-C), 1634(C=N), 1259(C-N str.), 993(N-N), 742(C-S-C) cm<sup>-1</sup>.

**(4d):** IR, 3446(N-H), 3076, (C-H), 1497, 1594(C=C), 1257(C-O-C),1638(C=N), 1257(C-N), 994(N-N), 748(C-S-C) cm<sup>-1</sup>.

(4e): IR, 3442(N-H), 3079, 3025(C-H.), 2995(C-H), 2918, 2810(C-H), 1494, 1516, 1593(C=C), 1254(C-O-C), 1630(C=N), 1254 (C-N), 995(N-N), 753(C-S-C) cm<sup>-1</sup>. (41): IR, 3441(N-H), 3075, 3021(C-H), 1375(C-H), 1499, 1518,1593(C=C), 1256(C-O-C), 1078(C-O-C), 1632(C=N),1256(C-N),996(N-N), 745(C-S-C) cm<sup>-1</sup>. **(4g):** IR, 3447(N-H), 3078, 3023(C-H), 1498, 1513, 1592 (C=C1254(C-O-C), 1634(C=N), 1254(C-N str.), 994(N-N), 756(C-S-C) cm<sup>-1</sup>.





# Antimicrobial Activities: Procedure (cup plate agar disc-diffusion method)

Solutions of tested compounds (4a-g) were prepared by dissolving calculated amount of every yielded product in DMSO to give final concentration of 31-1000µg/mL. Petri plates and agar solution are sterilized in autoclave. Petri plate is arranged by pouring agar solution. Bacterial culture was inoculated on fresh nutrient broth and further diluted with water 0.1 mL of diluted culture was banquet over nutrient agar in plate. Sterilized Whatmann paper circles (6mm) were soaked in different tested compoundsand dried at room temperature then applied on petri plate and incubated at 37°C for 24 h zone of inhibition was measured in all direction in mm and taken as mean. Consequence was correlated with reference drug Chloramphenicol.

# **RESULT AND DISCUSSION**

All novel synthesized compounds 4a-g

has been corroborated by the spectroscopic examination such as FT-IR, 1HNMR, mass spectra and <sup>13</sup>CNMR. For every final and intermediate product solubility and melting point was determined. The synthesis of target compound 6-amino triazolothiadiazole derivatives 4a-g was carried out by cyclocondensation reaction of compound (3) with aryl isothiocyanate in DMF and potassium carbonate. The synthetic protocol has been outline in scheme1. The FT-IR result of compounds 4a-g, revealed the disappearance of absorption bands due to -SH and -NH<sub>2</sub> stretching frequencies of initial compounds 3 and appearance of broad band at 3440 cm<sup>-1</sup> for NH stretch. The new band which appeared at 1636 cm<sup>-1</sup> region is recognized to stretching frequency of C=N group of the thiadiazole ring further more absorption band at 749 cm<sup>-1</sup> shows C-S-C stretch, undoubtedly indicated the cyclisation of compounds 3 and arylisothiocyanate in the presence of potassium carbonate to form triazolothiadiazole fused ring 4a.

s s s s s s	Physics M.F Colour M.Pt Yield Rf. Recys. S M.F Colour M.Pt	al Data : : : : : : : :	a(4a-g) C <sub>20</sub> H <sub>16</sub> BrN <sub>7</sub> OS White solid 275℃ 85% 0.73 Ethanol C <sub>20</sub> H <sub>16</sub> CIN <sub>7</sub> S	%N Found (Calcd.) N, 17.6 (N, 17.02)	%S Found (Calcd.) S, 5.78 (S, 5.6)
s - D - Br - Br	Colour M.Pt Yield Rf. Recys. S M.F Colour	::	White solid 275ºC 85% 0.73 Ethanol	N, 17.6	S, 5.78
s D-Br	Colour M.Pt Yield Rf. Recys. S M.F Colour	::	White solid 275ºC 85% 0.73 Ethanol		
s Бран	M.Pt Yield Rf. Recys. S M.F Colour	::	275ºC 85% 0.73 Ethanol		
s	Yield Rf. Recys. S M.F Colour	: : :	85% 0.73 Ethanol	(N, 17.02)	(S. 5.6)
Б Б Б С Д сі	Rf. Recys. S M.F Colour	:	0.73 Ethanol	(N, 17.02)	(S. 5.6)
s C C I	Recys. S M.F Colour	:	Ethanol	(N, 17.02)	(S. 5.6)
s D-a	M.F Colour	:			(0, 0.0)
s D-ci	Colour		CHCINS		
s D-a			26 16 70		
- Cl-ci	M Pt	•	White solid	N, 19.23	S, 6.29
-CI		:	260°C		
	Yield	:	87%		
	Rf.	:	0.65	(N, 19.10)	(S, 6.25)
-	Recys. S	:	Ethanol		
•	M.F	:	C26H16CIN7S		
s	Colour	:	White solid	N, 19.23	S, 6.29
ଁ୍ମ	M.Pt	:	264°C		
N-(	Yield	:	83%		
	Rf.	:	0.7	(N, 19.16)	(S, 6.21)
	Recys. S	:	Ethanol		
N II	M.F	:	C <sub>27</sub> H <sub>19</sub> N <sub>7</sub> OS		
٩	Colour	:	White solid	N, 20.03	S, 6.55
₹ <sup>2</sup> \	M.Pt	:	270°C		
HN -	Yield	:	78%		
	Rf.	:	0.68	(N, 19.98)	(S, 6.47)
	Recys. S	:	Ethanol		
N U	M.F	:	C <sub>27</sub> H <sub>10</sub> N <sub>7</sub> OS		
~	Colour	:	White solid	N, 20.03	S, 6.55
<i>≼</i> " ।	M.Pt	:	258°C		
		:	88%		
	Rf.	:	0.58	(N, 19.78)	(S, 6.50)
	Recys. S	:	Ethanol		
	M.F	:			
	Colour	:		N, 20.03	S, 6.55
s		:			
~~~~		:			
N-{ -]				(N. 20.01)	(S, 6.52)
				( ) )	(-, )
n 1				N. 20.62	S, 6.74
~				, _0.02	2, 01
≈<"					
HN	Rf.		01/0		
11 I	BI	:	0.76	(N, 20.58)	(S, 6.68)
		Recys. S M.F Colour M.Pt Yield Rf. Recys. S M.F Colour M.Pt Yield Rf. Recys. S M.F Colour Rf. Recys. S M.F Colour S M.F Colour Rf. Recys. S M.F Colour S M.F Colour S M.F Colour Rf. Recys. S M.F Colour S M.F Colour S M.F Colour S M.F Colour M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Yield Rf. Rf. Recys. S M.F Yield Rf. Rf. Recys. S M.F Yield Rf. Rf. Rf. Rf. Rccys. S M.F Colour S M.F Colour S M.F Yield Rf. S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F Colour S M.F	Recys. S : M.F : Colour : M.Pt : Yield : Rf. : Recys. S : M.F : Colour : M.F : Colour : M.F : Colour : M.F : Colour : M.F : Colour : M.F : Colour : M.F : Recys. S : M.F : Recys. S : M.F : Recys. S : M.F : Colour : M.F : Colour : M.F : Colour : M.F : Colour : M.F : Colour : M.F : Recys. S : M.F : Colour : M.F : Recys. S : M.F : Colour : M.F : M.F : Colour : M.F : M.F : Colour : M.F : M.F : M.F : M.F : Colour : M.F : M.F : Colour : M.F : M.F : Colour : M.F : M.F : Colour : Co	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

Table 1: Analytical and Physical data of 6-Amino Triazolothiadiazoles (4a-g)

<sup>1</sup>HNMR of 4a indicated disappearance of proton peak for –SH and –NH<sub>2</sub> and occurrence of singlet at  $\delta$  14.56 ppm shows the existence of aromatic secondary NH group, and another singlet at  $\delta$  6.59 ppm shows existence of one proton at C4 of pyrazole ring. Apart from above signal rest of the signals were observed in the aromatic region. <sup>13</sup>C signals of 4a recorded a singlet at  $\delta$  161.54 ppm due to and another signal at  $\delta$  171.2 ppm due to C8 carbon of triazolothiadiazole that reveals expected cyclization, while other signals of <sup>13</sup>CNMR spectra of 4a were obtained at predicted chemical shifts values. Moreover the % of C, H, N and S were establish to be 56.40, 2.98, 17.74, 5.70 respectively, which also indicates that it is in good conformity with molecular formulae  $C_{26}H_{16}BrN_7OS$ . Base peak in GC-MS (m/z) at [M+2]<sup>+</sup> at 556 also supported the formation of target compound. The nonappearance of characteristic absorption peaks due to -NH<sub>2</sub> and -SH groups in 4a as earlier shown in 3 clearly confirmed its formation. The entire synthesized heterocyclic compounds 4a-g was assessed for their

in-vitro antimicrobial activity. Results obtained are summarised in the Table No. 2 and 3.

				Zone	of Inhibitio	n (mm)								
		Gram +ve							Gram –ve					
Compd.Code	•			P. vulgaris										
		Conc. (µg/mL)												
	1000	500	250	125	63.5	31	1000	500	250	125	63.5	31		
4a	24	21	19	18	17	14	26	23	20	18	16	12		
4b	22	20	19	18	16	14	24	22	21	19	17	15		
4c	20	19	17	15	14	12	25	21	19	16	15	12		
4d	21	18	16	15	13	11	22	20	18	15	14	13		
4e	25	21	19	17	16	15	23	22	21	19	16	14		
4f	24	21	20	19	18	16	27	25	22	18	15	13		
4g	22	20	19	17	16	14	23	21	20	17	15	14		
DMSO	-	-	-	-	-	-	-	-	-	-	-	-		
Std. Drug	24	22	20	19	17	15	28	24	20	17	16	13		

Table 2: Antibacterial screening of (4a-i)

### Table 3: Antibacterial screening of (4a-i)

				Z	one of Inhi	bition (m	m)					
Compd. Code	Э				Gram	-ve						
			E. (	S.typhi								
Conc. (µg/mL)												
	1000	500	250	125	63.5	31	1000	500	250	125	63.5	31
4a	26	24	22	19	17	15	16	15	13	10	09	07
4b	23	21	20	18	16	14	14	12	11	09	08	06
4c	24	22	19	17	15	13	15	13	10	08	07	05
4d	26	23	22	20	18	15	12	10	09	07	06	04
4e	25	24	21	18	17	14	15	14	12	11	09	07
4f	24	23	20	17	16	13	13	12	10	08	07	05
4g	20	19	17	16	14	12	14	13	11	10	08	06
DMSO	-	-	-	-	-	-	-	-	-	-	-	-
Std. Drug	26	24	23	21	17	14	17	15	12	11	09	08

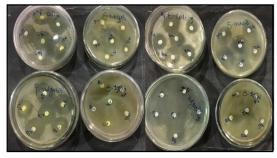


Fig. 1. Zone of Inhibition in mm for 4a and 4e at different conc. and bacterial strains

## CONCLUSION

In conclusion, highly efficient and facile synthesis of novel series of 6-amino thiadiazole (4a-g) derivatives is described. The presented series of compounds were synthesized in excellent yields without any additional reagents or catalyst. The structure and purity of innovative compounds obtained was established by spectroscopic study and chemical assessment. Among the synthesized compounds maximum compounds exhibited reasonable to good activities on selected strains *S. aureus, E. coli* and *P. vulgaris* while poor activity was assessed for *S. typhi.* 

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### Conflict of interest

The authors announce that there is no conflict of interest.

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