

ORIENTAL JOURNAL OF CHEMISTRY

An International Open Free Access, Peer Reviewed Research Journal

ISSN: 0970-020 X CODEN: OJCHEG 2018, Vol. 34, No.(4): Pg. 2137-2143

www.orientjchem.org

Estimation of Lattice Constants and Band Gaps of Group-III Nitrides using Local and Semi Local Functionals

SANDEEP ARORA^{1*}, DHARAMVIR SINGH AHLAWAT^{1*} and DHARAMBIR SINGH²

¹Department of Physics, Chaudhary Devi Lal University, Sirsa (1250 55), India. ²Department of Physics, M M University, Mullana, India. *Corresponding author E-mail: sndp.arora@rediffmail.com

http://dx.doi.org/10.13005/ojc/3404055

(Received: June 08, 2018; Accepted: July 01, 2018)

ABSTRACT

We performed the optimization of lattice constants of Group-III nitrides (InN, AIN, GaN) in wurtzite and Zinc blende structures using various semilocal exchange correlation functional in generalized gradient approximations (GGA) namely PBE, WC, PBEsol in addition to local density approximation (LDA) functional. We used these optimized lattice parameters to predict the band gap values using modified Becke Johnson exchange potential with original and improved parameterization as suggested by David Koller for semiconductors having band gap values below 7eV. Among the different functionals considered, PBEsol optimize the lattice parameters with smallest mean error (0.00639 A) relative to experimental values, while WC approximation with a slightly greater value of mean error (0.00513 Å). It is shown that mBJLDA approximation improves the band gap for the materials studied when compared with LDA and GGA results. It is also shown that LDA optimized parameters with mBJLDA approximation, which leads to mean error of 0.162 eV reproduces the experimental band gap in most efficient way.

Keywords: Group-III nitrides, Band Gap, LDA, GGA, mBJLDA.

INTRODUCTION

Group III- nitride semiconducting materials form the host material system for the fabrication of optoelectronic devices working in wide spectrum from infrared region to ultraviolet region¹⁻³. Alloying the other group III or V elements in host material tune the optical and structural properties for desired potential applications. There are reports, which show the nonlinear variation in structural properties and strong shrink in band gap of InNBi alloy with composition of Bi4. For both wurtzite and zinc blende phases, the band gap varies from 0.8 eV to 3.5 eV of InGaN alloys with Ga composition5. For designing new materials and tailoring them for desired optical applications, we need a fast and accurate determination of their band gap values. The calculation of properties of proposed materials using computational methods provides an efficient way for the experimentalist to predict band gap prior to their



This is an Open Access article licensed under a Creative Commons Attribution-Non Commercial-Share Alike 4.0 International License (https://creativecommons.org/licenses/by-nc-sa/4.0/), which permits unrestricted BY NC SA Non Commercial use, distribution and reproduction in any medium, provided the original work is properly cited.

using mBJLDA exchange correlation potential with original and improved version suggested for semiconductors having band gap below 7 eV.

METHOD

All the calculations were carried out using DFT based all electron full potential linearized augmented plane wave and local orbital (FPLAPW + Io) method as implemented in wein2k code.²⁶ Three GGA functionals i.e. PBE, WC and PBEsol were considered for taking into account the exchange correlation effect. For improving the band gaps as obtained with local and semilocal functional, we employed mBJ exchange potential in combination with LDA correlation¹⁴. The mBJ exchange potential is defined by equation (4) of reference¹⁶.

$$v_x^{TB-mBJ}(\mathbf{r}) = c v_x^{BR}(\mathbf{r}) + (3c-2)\frac{1}{\pi}\sqrt{\frac{5}{12}}\sqrt{\frac{2t(\mathbf{r})}{\rho(\mathbf{r})}}$$
 (1)

Where ρ being the electron density, t is the kinetic energy density and $v_x^{\,\rm TB-mBJ}$ is the Becke-Russel potential and parameter c is given by

C =

$$g = \frac{1}{V_{cell}} \int_{cell} \frac{1}{2} \left(\frac{|\nabla \rho \uparrow (\boldsymbol{r})|}{\rho \downarrow (\boldsymbol{r})} + \frac{|\nabla \rho \downarrow (\boldsymbol{r})|}{\rho \uparrow (\boldsymbol{r})} \right) d^3 r \quad (3)$$

In equation (2) A, B and e are the parameters whose values are fitted according to the experimental values. The parameters suggested in original and improved form of mBJ approximation are shown in Table 1.

Table 1: The values of parameters A, B and e used for calculations of 'c' in [Eq. (2)]

	А	В	е	
Poriginal	-0012	1.023	0.5	
Psemiconctor	0.267	0.656	1	

In Table 1, $P_{original}$ and $P_{semiconductor}$ are the parameterization used for calculation of 'c' in original and improved form of mBJ approximations as suggested by Tran & Blaha¹⁴ and D. Koller respectively¹⁶. All the calculations were converged with respect to number of k- points in Brillouin zone and the value of $R_{mt} K_{max}$ (between 8 and 10) defining the size of basis set used. For analyzing results, the following statistical quantities have been

synthesis. Density functional theory based methods are tools of interest to compute these properties due to their low computational cost. The standard DFT local LDA and semi local GGA exchange correlation functional suffered from self-interaction problem, which leads to the underestimation of band gaps relative to experimental values⁶⁻⁸. Therefore, alternative methods are used to predict accurate band gaps of materials. Quasi particle self-consistent GW method gives best theoretical description of electronic properties of materials⁹⁻¹¹. However due to their large computational cost these methods usually not applied to large systems. The hybrid semi local exchange with nonlocal Fock exchange¹²⁻¹³ provides the band gaps comparable to GW results at a relatively low cost still much expensive as compared to standard local and semi local exchange functional. Recently Tran and Blaha proposed a multiplicative exchange potential¹⁵ modifying over original Becke Johnson potential¹⁵ which predict the band gap with accuracy comparable to GW method but at cost in the range of semi local functional. We used the TB-mBJ potential to calculate theoretical band gaps of Group III nitrides in wurtzite and Zinc blende structures using original and improved parameterization for empirical parameter 'c' in equation(4) of reference¹⁶.

A small deviation in lattice parameters may result in significant variation in band gaps for same semiconductor¹⁷. Therefore, while calculating band gap the optimization of lattice parameters should be done in an efficient way, which in turn depends upon the selection of proper exchange correlation functional. TB-mBJ is an exchange potential and cannot be obtained as a derivative of exchange correlation energy. Thus, lattice parameters have to be optimizing with LDA or GGA approximation before calculating the band structure with TB-mBJ potential¹⁴. In Kohn Sham (KS) equations based density functional theory^{18,19} the exact exchange correlation functional is not known. Thus, it is of utmost importance to select a proper functional for the problem in hand.

In the present work, we performed the optimization of lattice parameters of materials under study of group-III nitrides using four approximations i.e. LDA^{20,21}, PBE^{22,23}, WC²⁴ and PBEsol²⁵. With these lattice parameters, the band gaps were calculated

(2)

used: mean error (ME), mean absolute error (MAE), mean relative error (MRE, in percentage), and mean absolute relative error (MARE, in percentage).

RESULTS AND DISCUSSION

Equilibrium Lattice constant

At ambient conditions III-nitride semiconductors prefer to crystallize in hexagonal structure with space group $P6_3mc^{27}$. However, these materials can also be grown in zinc blende structure

using experimental techniques like molecular beam epitaxy²⁷. The wurtzite structure can be supposed to be the interpenetration of the two hexagonal structures with relative displacement along c axis of unit cell while zinc blende structure can be imagined as the interpenetration of two face centered cubic structures with one displaced relative to other by one fourth along body diagonal of unit cell. The wurtzite structure is characterized by three parameters namely a, c/a and u while the zinc blende unit cell by only one parameter a.

2139

Table 2: Equilibrium lattice parameters (in Å) of III-N semiconductors,
calculated with LDA, PBE, WC and PBEsol along with
Experimental values ²⁸

Materia	l	LDA	PBE	WC	PBEsol	Expt.
wzInN	а	3.5073	3.5852	3.5421	3.5408	3.545
	С	5.6689	5.7916	5.7226	5.7191	5.703
zbInN	а	4.9461	5.0545	4.9942	4.9921	4.98
wzAIN	а	3.0908	3.132	3.1142	3.1151	3.112
	С	4.9478	5.0219	4.9874	4.987	4.982
zbAlN	а	4.3467	4.4064	4.38	4.3809	4.38
wzGaN	а	3.1593	3.2219	3.1893	3.1884	3.189
	С	5.1497	5.248	5.1964	5.1939	5.185
zbGaN	а	4.4619	4.5512	4.5073	4.5058	4.5
	ME	-0.0341	0.048	0.00639	0.00513	
	MAE	0.0341	0.048	0.00703	0.0063	
	MRE(%)	-0.799	1.069	0.124	0.101	
	MARE(%)	0.799	1.069	0.142	0.131	

The optimized lattice parameters obtained with different approximations are shown in Table 2 with respective errors ME, MAE, MRE and MARE in various calculations written at bottom of the table. The calculated results are also shown graphically in Fig. 1 in which the inclined straight line represents the experimental values while calculated parameters by different symbols.



Fig. 1. Calculated versus Experimental lattice parameters of III-N (InN, AIN, GaN) using various functionals

The graph shows that the PBEsol lattice constants lie closest to experimental values as compared to other results. The equilibrium LDA lattice constant underestimate while PBE overestimate the experimental values with ME of -0.0341 Å and 0.048 Å respectively as reported earlier²⁹. The functional WC and PBEsol yields the lattice constants much closer to experimental values as suggested by lowest values of statistical quantities written at the bottom. With small differences in ME, MAE, MRE and MARE for WC and PBEsol approximations, both of these functional are equally applicable for optimization of lattice constant for Group-III nitrides. Table 3 shows a comparison of our calculated WC and PBEsol optimized lattice constants with hybrid

density functional results^{30,31} and lattice parameters optimized with OPT functional as given in recently proposed database³². The comparison concludes that PBEsol reproduces the experimental values in better way, supported by minimum values of ME and MAE of 0.0063 Å and 0.0073 Å when compared with present WC, hybrid functional used in reference³¹ with ME, MAE of -0.0073 Å, 0.0091 Å and OPT functional used in reference³² with ME and MAE of 0.041 for InN, AIN, GaN in wurtzite and zinc blende structures respectively. The hybrid functional with optimized α parameter used in reference³⁰ underestimate the lattice constants for III-N semiconductors in zinc blende phase with ME of -0.018 Å while the present PBEsol results overestimate with ME of 0.0063 Å.

2140

Table 3: Comparison of WC and PBEsol optimized lattice constants (Å) with hybrid and OPT functional results

		WC	PBEsol	Hybrid	OptB88-vdWc	Expt.
wzInN	а	3.5421	3.5408	3.542 ^b	3.572	3.545
	с	5.7226	5.7191	5.711 [♭]	5.784	5.703
zblnN	а	4.9942	4.9921	4.956ª, 4.988 ^b	5.039	4.98
wzAIN	а	3.1142	3.1151	3.103⁵, 3.064ª	3.13	3.112
	С	4.9874	4.987	4.970 ^b ,4.908 ^d	5.021	4.982
zbAlN	а	4.38	4.3809	4.367ª,4.363 ^b	4.404	4.38
wzGaN	а	3.1893	3.1884	3.180 [♭] ,3.232₫	3.211	3.189
	с	5.1964	5.1939	5.172 ^b ,5.268 ^d	5.24	5.185
zbGaN	а	4.5073	4.5058	$4.483^{a}, 4.489^{b}$	4.541	4.5

All the experimental values are taken from ref.²⁸. ^a from ref.³⁰.^b from ref.³¹.^c from ref.³².^d from ref.³³.

Band gap

The calculated band gap values for III-nitrides in wurtzite and zinc blende structures are shown in Table 4 along with the various errors namely ME, MAE, MRE, MARE given at bottom of the Table. The results are also shown graphically in Fig. 2. All the local and semilocal functionals strongly underestimate the band gaps with mean error of about - 1.5 eV, mean relative error of about -60%. This is due to self interaction effect suffered by the respective functional used. The application of mBJ exchange potential in association with LDA correlation as suggested by Tran Blaha¹⁴ improves the band gaps leading to a decrease in ME and MARE to values of order of -0.2eV and 10% respectively. A further slight improvement in band

gap is obtained with improved version¹⁶ of mBJLDA approximation leading to the reduction in various errors as shown at the bottom of the Table 4 except for the results at LDA optimized lattice constants where the errors increase. Further, in case of wz InN the improved mBJLDA approximation increases the deviation from experimental values as compared to original mBJLDA version. The best agreement with experimental values is obtained with mBJLDA approximation at LDA optimized parameters with MAE and MARE of 0.162eV and 7.41% respectively. However the results obtained with improved mBJLDA at WC and PBEsol optimized lattice constants also compete in accuracy with mBJLDA + LDA as suggested by MAE and MARE values of 0.198eV, 0.199eV and 9.81%, 9.99% respectively.

Table 4: Calcalated band gap (eV) with local, semilocal functional, original mBJLDA and improved mBJLDA (mBJLDASC) for semiconductors. Experimental values are taken from reference²⁸

	LDA	mBJ LDA	mBJ LDA (SC)	PBE	mBJ LDA	mBJ LDA (SC)	WC	mBJ LDA	mBJ LDA (SC)	PBEsol	mBJ LDA	mBJ LDA (SC)	Expt.
wzInN	0	0.978	1.169	0	0.739	0.913	0	0.866	1.05	0	0.87	1.052	0.78
zbInN	0	0.758	0.939	0	0.535	0.696	0	0.657	0.827	0	0.67	0.839	0.78
wzAIN	4.397	5.759	5.926	4.029	5.391	5.544	4.104	5.55	5.715	4.096	5.546	5.705	6.19
zbAIN	3.242	4.913	5.104	3.311	4.846	5.016	3.22	4.871	5.05	3.188	4.851	5.03	4.9
wzGaN	2.089	3.337	3.566	1.707	2.954	3.171	1.825	3.152	3.376	1.857	3.162	3.386	3.503
zbGaN	1.93	3.148	3.36	1.546	2.768	2.975	1.654	2.952	3.163	1.682	2.959	3.168	3.29
ME	-1.297	0.092	0.104	-1.475	-0.368	-0.188	-1.44	0.232	-0.0437	1.437	-0.231	0.0438	
MAE	1.297	0.162	0.192	1.475	0.368	0.271	1.44	0.261	0.199	1.437	0.261	0.198	
MRE	-57.42	-7.33	12.33	-61.94	-13.7	-3.47	-60.94	-5.99	4.76	-60.77	-0.925	6.15	
(%)													
MARE	57.42	7.41	13.75	61.94	13.7	9.95	60.94	9.67	9.81	60.77	9.47	9.997	
(%)													



Fig. 2. Calculated versus Experimental band gaps of III-N (InN, AIN, GaN) using mBJLDA and mBJLDA (PSC) at various optimizations

On comparing our results with other mBJLDA band gaps we found an improvement as shown in Table 5, this is due to the reason that the reported calculations were performed at experimental lattice constants while our calculations are at optimized parameters which is obvious as suggested by J.A. Camargo that the band gaps calculated with experimental lattice constants show greater deviation from experimental band gap values using mBJLDA potential¹⁴.

As GW and hybrid density functional methods are known for the best theoretical methods to describe the electronic structure of solids so it is necessary to compare our results with above said methods, in addition, the band gaps calculated at OPT optimized lattice parameters as given in recently launched database³² are also shown in Table 5 for comparison. The MAE of 0.198eV for mBJLDA band gaps suggest that this approximation have the accuracy of the order of hybrid density functional and GW methods having the MAE of 0.105 eV and 0.189 eV respectively as implemented in references^{35,36} respectively but at relatively lower computational cost. For zinc blende III-N semiconductors, mBJLDA with PBEsol optimized geometry gives band gaps in better agreement with experimental values with MAE of 0.104 eV as compared to 0.203 eV when calculated with hybrid density functional method as reported in reference³⁰ by Bastos *et. al.*, The band gaps of III-N semiconductors, reported in JARVIS-DFT database calculated with mBJ exchange potential at optB88vdW functional (OPT) are found to underestimate the experimental gaps with ME, MAE, MRE and MARE -0.353eV, 0.353eV, -11.65%, 11.65% respectively³². While our calculated mBJLDA band gaps at PBEsol optimized lattice parameters predict the band gaps with respective values of statistical quantities as 0.0438eV, 0.198eV, 6.15%, 10%.

mBJLDA	mBJLDA	mBJLDA ³⁴ SC+PBEsol	HSEα ³⁰	B3PW91 ³⁵	GW ³⁶ m	BJLDA+OPT ³²	Expt.	
wzInN zbInN wzAIN zbAIN wzGaN zbGaN	0.978 0.758 5.759 4.913 3.337 3.148	1.052 0.839 5.705 5.03 3.386 3.168	0.89 5.51 4.88 3.17 2.85	0.674 5.383 3.312	0.85 6.05 4.94 3.44 3.08	0.61 0.38 6.4 5.19 3.45 3.3	0.76 0.58 5.2 4.8 3.084 2.9	0.78 0.78 6.19 4.9 3.503 3.29

Table 5: Comparison of band gap ((eV) with other reported results
-----------------------------------	----------------------------------

CONCLUSION

We have made a systematic investigation, by the use of various local and semi local functionals, for optimization of lattice constants and prediction of band gaps of group-III nitrides (InN, AIN, GaN) in wurtzite and zinc blende structures. Importantly, the PBEsol exchange correlation functional is found to provide lattice constant closest to experimental values shown by smallest M(A)E, M(A)RE of 0.00513 Å (0.0063 Å) and 0.101%(0.131%) with WC approximation also competing it with corresponding errors 0.00639 Å (0.00703 Å) and 0.124% (0.142%) for the materials studied. Further, these approximations have also predicted the band gap values closer to their experimental values when combined with mBJLDA potential with parameterization suggested for semiconductors. However, the best agreement for band gap values with experiment is obtained with mBJLDA +LDA approximation.

ACKNOLEDGEMENT

The authors thank Fabien Tran for valuable discussion through wien2k official mailing list.

REFERENCES

- 1. Jani, O.; Ferguson, I.; Honsberg, C. & Kurtz, S. Appl. Phys. Lett., **2007**, *91*, 132117.
- Dahal, R., Pantha, B., Li, J., Lin, J. & Jiang, H. Appl.Phys. Lett., 2009, 94, 063505.
- 3. Nakamura, S. & Fasol, G. Springer **1997** (ed. Nakamura, S. & Fasol, G.)50.
- Lu ,P.; Liang,D.; Chen,Y.; Zhang, C.; Quhe, R.; Wang S. Scientific Reports 2008, Article number : 10594.
- 5. Gan, C. K.; Feng, Y. P.; Srolovitz D. *J. Phys. Rev.* B., **2006**, *73*, 235214.
- 6. Perdew, J. P.; Zunger, A.; *Phys. Rev.*, B **1981**, *23*, 5048.
- 7. Perdew, J..P.; Levy, M.; *Phys. Rev. Lett.*, **1983**, *51*, 1884.

- Jones, R.O.; Gunnarsson, O. *Rev. Mod. Phys.*, 1989, *61*, 689.
- 9. Hedin, L. Phys. Rev., **1965**, *139*, A796.
- Hybersten, M. S.; Louie, S. G. Phys. Rev. B : Condens. Matter Mater. Phys., 1986, 34, 5390.
- 11. Shishkin, M. S.; Marsman, M.; Kreese, G. *Phys. Rev. Lett.*, **2007**, *99*, 246403.
- 12. Perdew, J. P.; Erhzerhof, Burke, K. *J. Chem. Phys.*, **1996**, *105*, 9982.
- 13. Adamo, C.; Barone, V. *J. Chem. Phys.*, **1999**, *110*, 6158.
- 14. Tran, F.; Blaha, P. *Phys. Rev. Lett.*, **2009**, *102*, 226401.
- Becke, A. D.; Johnson, E. R. J. Chem. Phys., 2006, 124, 221101.

- Koller. D.; Tran, F.; Blaha, P. *Phys. Rev.*, B 2012, *85*, 155109.
- 17. Camargo, J. A.; Martinez, Baquero, R. *Phys. Rev.*, B **2012**, *86*, 195106.
- 18. Hohnberg, P.; Kohn, W. *Phys. Rev.*, **1964**, *136*, B864.
- Kohn, W.; Sham, L. *J. Phys. Rev.*, **1965**, *140*, A1133.
- 20. Dirac, P.A.M. Proc. Camb. *Philos. Soc.*, **1930**, *26*, 376.
- 21. Perdew, J. P.; Wang, Y. *Phys. Rev.*, B **1992**, *45*, 13244.
- Perdew, J. P.; Chevary, J. A.; Vosko, S. H.; Jackson, K.A.; Pederson, M. R.; Singh, D. J.; Fiolhais, C. *Phys. Rev.*, B **1992**, *46*, 6671.
- 23. Perdew, J. P.; Burke, K.; Ernzerhof, M. *Phys, Rev. Lett.*, **1996**, *77*, 3865.
- 24. Wu, Z.; Cohen, R. E. *Phys. Rev.*, B **2006**, *73*, 235116.
- Perdew, J. P.; Ruzsinszky, A.; Csonka, G. I.;
 Vydrov, O. A.; Scuseria, G. E.; Constantin,
 L.A.; Zhou, X.; Burke, K. *Phys. Rev. Lett.*,
 2008, *100*, 136406.
- 26 Blaha, P.; Schwarz, K.; Madsen, G. K. H.; Kvasnicka, D.; Lnitz, J. Vienna University of

Technology, Austria, 2001.

- Morkoc, H. Handbook of Nitride Semiconductors and Devices, WILEX-VCH Verleg, GmbH & Co. KGaA, Weinhein., 2008, 1, 2047.
- Vurgaftman, I.; Meyer, J. R. J. Appl. Phys., 2003, 94, 3675.
- 29. Hass, P.; Tran, F.; Blaha, P. *Phys. Rev.*, B **2009**, *79*, 085104.
- Bastos, C. M. O.; Sabino, F. P.; Sipehi, G. M.; Da Silva, L. F. J. Appl. Phys., 2018, 123, 065702.
- Caro, M. A.; Schulz, S.; O'Reilly, E. P. *Phys. Rev.*, B 2012, *86*, 014117.
- Choudhary, K.; Zhang, Q.; Reid, A. C. E.; Chowdhary, S.; Nguyan, N. V.; Trautt, Z.; Newrock, M. W.; Congo, F. X.; Tavazza, F. Scientific Data., 2018, 05.
- Qin, L.; Duan, Y.; Shi, H.; Shi, I.; Tang, G. J. Phys. : Condens. Matter., 2013, 25, 045801.
- 34 Tran, F.; Blaha, P. J. Phys. Chem., 2017, A121, 3318.
- 35. Crowley, J. M.; Tahir Kheli, J.; Goddard, W. A. *J. Phys. Chem. Lett.*, **2016**, *7*, 1198.
- 36. Deguchi, D.; Sato, K.; Kino, H.; Kotani, T. Jpn. *J. Appl. Phys.*, **2016**, *55*, 051201.