

Synthesis, characterization and biological activities of some new acid hydrazones

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ABSTRACT

A series of new acidhydrazones have been synthesised by the reaction of 2,3-dichloroanilido acetohydrazide with various carbonyl compounds in 58 to 92% yield. Newly synthesized compounds (1,3,4,7,8,9,12,13,15and16) have been tested for their anti-bacterial activity against gram positive bacteria *S.albus*, *S.aureus* and *Gram negative* bacteria *E.coli* and *Pseudomonas piosineus*. The compound 1,3,12,13 and 15 shown significant activity and compound 4,7,8 and 9 have shown moderate activity. The same compounds were tested for their anti-fungal activity against *Candida albicans*, *Aspergillus niger* and *Alternaria alternate* at concentration of 30 mg/ml using sabouraud dextrose agar media. Compounds 12,13 and 15 were found to be moderately active against *candida albicans* and *aspergillus niger*. All the other compounds did not show significant activity against the fungi at the concentration used.

Key words: Malonicester, Acidhydrazide, Acidhydrazones, synthesis, Characterization, and Biological Activities.

INTRODUCTION

Hydrazones possessing an azometine - NHN=CH- Proton constitute an important class of compounds for new drug development. Therefore, many researchers have synthesized these compounds as target structures and evaluated their biological activities. Acid hydrazides have frequently been investigated for testing their potentiality as tuberculostats¹⁻⁸. Hydrazides and their condensation products have displayed diverse range of biological properties such as bacteriocidal⁹⁻¹⁰, anti-fungal¹¹, anti-convulsant¹²⁻¹⁵, anti-helminthic¹⁶, anti-tumor¹⁷⁻²⁰, anti-leprotic²¹, anti-malarial²²⁻²³, anti-cancer²⁴⁻³¹, anti-depressant³², anti-HIV³³, analgesic-anti-inflammatory³⁴, leishmanicidal³⁵, vasodilator activities³⁶.

EXPERIMENTAL

All chemicals used were of A.R. grade (either of B.D.H. or Excel-R or extra pure E. Merk quality). The structure of the compounds were determined by elemental analysis, IR and NMR spectral data. IR spectra (KBr) are recorded on a perkin-Elmer 283 spectrophotometer. NMR

spectra (CDCl_3) are recorded on varian EM 360 L spectrophotometer. Melting point of the compounds are determined in open capillary tubes and are uncorrected. Purity of the compounds is checked on T.L.C. using silica gel-G. Elemental analysis is performed on Carlo-Erba 1108 analyser.

General procedure

Preparation of Ethyl-2-(2,3-dichloroanilido)ethanoate [1]

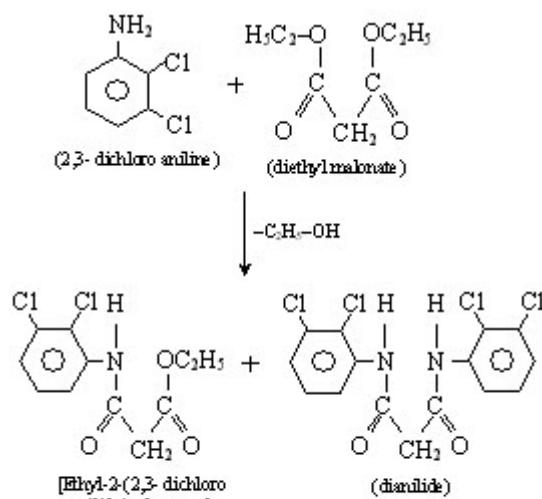
A mixture of 2,3-dichloro aniline (5 ml) and diethyl malonate (10ml) was refluxed for 50-55 minutes in a 100 ml r.b. flask fitted with an air condenser of such a length (14²) that ethanol formed escaped and diethyl malonate flowed back in to the flask. Contents were cooled, 30 ml ethanol was added and kept over night, well precipitate found. It was filter under suction and purified by recrystallisation from ethanol.

IR Absorption band (cm⁻¹)

3150 (N-H stretching), 1665–1660 (C=O Ketone), 1090 (C-Cl Stretching), 760–755 (Di substituted benzene), NMR spectra (d Me₂CO), 1.1–1.2 (3 H, t, CH₃), 2.21 (2 H, s, CH₂), 4.0–4.22 (2 H, q, CH₂), 7.0–7.21 (4 H, m, ArH).

Preparation of Ethyl-2-(2,3-dichloro anilido)acetohydrazide [2]

Ethyl-2-(2,3-dichloroanilido) ethanoate [.02 mol (5.52 gm)] was dissolved in rectified spirit in a small 3 neck r.b. flask kept on ice bath and set-up mechanical stirrer. Hydrazine hydrate (80%, 13 ml) was added by dropping funnel slowly drop by drop.



*colour – light cream
*MP – 90°C
*MF – C₁₁H₁₂Cl₂N₂O₂
*Yield – 68%

*MF – C₁₁H₁₂Cl₂N₂O₂
*colour – dark cream
*MP – 181°C

Scheme 1.

The contents were stirred for 15-20 minutes. There were evolution of heat and reaction was spontaneous after 20 minutes, solid was filtered under suction and recrystallised from ethanol, then we get silver white crystals in good yield.

IR Absorption band (cm⁻¹)

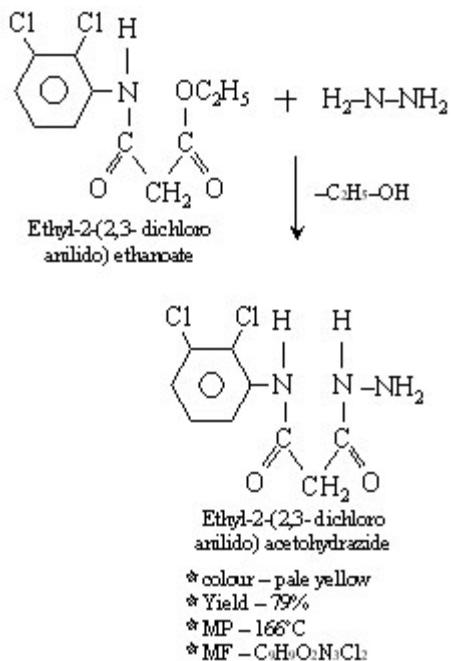
3160 (N-H stretching), 1660 (C=O Ketone), 1595, 1520, 1450 (C=C ring stretching), 760–755 (2, 3 di substituted benzene), NMR spectra (d DMSO), 2.25 (2 H, s, CH₂), 3.15 (3 H, s, CH₃), 4.12–4.31 (1 H, t, NH), 6.95–7.2 (3 H, m, ArH).

Synthesis of new Acidhydrazones [3]

Ethyl-2-(2,3-dichloroanilido)acetohydrazide (.001 mol) and (.001 mol) of aromatic aldehyde or ketone dissolve in absolute alcohol and added 2-drops of conc. H₂SO₄ and stirred for 15 minutes. It was filtered under suction and recrystallised from hot ethanol. Synthetic strategy has been outlined in scheme I,II&III. Mechanism for the formation of acid hydrazones is given in chart-I.

IR Absorption band (cm⁻¹)

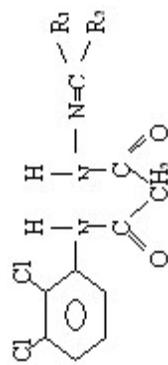
3150 (N-H stretching), 2960–2970 (C–H aliphatic), 1662–1660 (C=O Ketone), 785–778 (C–Cl Stretching), 760 (2,3-disubstituted benzene),



* colour – pale yellow
* Yield – 79%
* MP – 166°C
* MF – C₁₁H₁₂O₂N₂Cl₂

Scheme 2.

Table 1: Physical and analytical data of new compounds: Acid hydrazones derived from 2-(2,3-dichloroanilido) acetohydrazide



8.	3, 5 di chloro-2-hydroxy benzaldehyde	H Ph Cl (3) Cl (5)	OH(2) 214 ~	68 435	C ₁₆ H ₁₁ O ₃ N ₃ Cl ₄	White (44.11) (2.51)	2.52 (11.01) (9.64)	11.03 (32.64)	
9.	3-Nitro-6-hydroxy acetophenone	CH ₃ /Me Me	NO ₂ (3) OH (4) Me	220 49 194	C ₁₇ H ₁₄ O ₅ N ₄ Cl ₂ C ₁₂ H ₁₃ O ₂ N ₃ Cl ₂	Cream (48.00) (3.28)	3.29 47.68 (4.28)	18.82 13.17 (16.69)	
10.	Acetone	H	Ph-CI(2)	228 81	C ₁₆ H ₁₂ O ₂ N ₃ Cl ₃	White (49.92) (3.11)	3.12 49.93 (3.12)	10.92 (13.89) (23.49)	
11.	2-Chloro benzaldehyde	H	Ph-N-(CH ₂ -CH ₂ -CN) ₂	206 64	C ₂₂ H ₂₀ O ₂ N ₆ Cl ₂	Light brown	4.24 (56.04) (4.23)	6.79 (10.91) (27.68)	
12.	4-NN-Bis-2' cyano ethyl amino benzaldehyde	H	CH ₃ (2) N(CN ₂ -CH ₂ -CN) ₂	471 204	C ₂₃ H ₂₂ O ₂ N ₆ Cl ₂	Light brown	4.24 56.05 (6.78)	6.79 (17.83) (15.07)	
13.	2-Methyl-4-N-N-bis 2' cyano ethyl amino benzaldehyde	H	Ph-OCH ₂ -CH ₂ -CN(2)	86 ~	C ₂₃ H ₂₂ O ₃ N ₆ Cl ₂	Brown (56.89) (4.53)	4.53 (56.90) (4.53)	6.59 (17.31) (14.63)	
14.	2-Methoxy-4-N-N-bis 2' cyano ethyl amino benzaldehyde	H	OCN(2) N(CH ₂ -CH ₂ -CN) ₂	195 ~	C ₂₃ H ₂₂ O ₃ N ₆ Cl ₂	Brown (55.07) (4.38)	4.39 (9.57)	9.58 (16.75) (14.16)	
15.	Acetophenone	Me / CH ₃	Ph Ph-OH(2)	212 224	91 57	C ₁₇ H ₁₅ O ₂ N ₃ Cl ₂ C ₁₆ H ₁₃ O ₃ N ₃ Cl ₂	White White	4.12 (56.03) (4.11)	8.79 (8.78) (11.52) (19.48)
16.	Salicylaldehyde	H	Ph-OCH ₃ (2)	222	71	C ₁₇ H ₁₅ O ₃ N ₃ Cl ₂	Yellow (52.44) (3.54)	3.55 (13.10)	13.11 (11.46) (19.38)
17.	Anisic aldehyde	H	CH ₃	180	28	C ₂₃ H ₂₅ O ₂ N ₃ Cl ₂	Buff (53.67) (3.92)	5.60 (12.61)	12.63 (11.05) (18.68)
18.	β-Ionone	Me / CH3	CH ₃	446			7.17 (61.87) (5.59)	9.41 (7.14)	15.91 (9.39) (15.89)

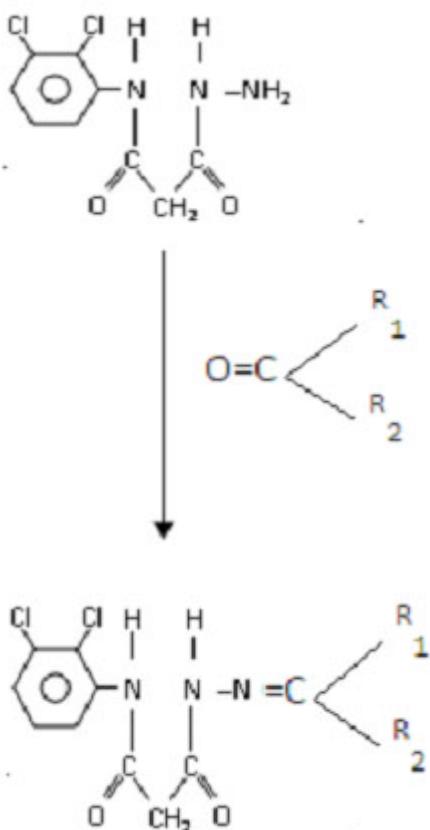
NMR spectra (d DMSO), 2.25 (2 H, s, CH₂), 4.21 (1 H, s, NH), 6.95–7.2 (10 H, m, ArH).

Biological evaluation

Anti-bacterial activity

Ten new acid hydrazones (1,3,4,7,8,9,12,13,15 and 16) were screened for their anti-bacterial activity against the gram positive

bacteria *S. albus*, *S. aureus* and gram negative bacteria *E. coli* and *pseudomonas piosineus* using cup-plate agar diffusion technique¹³ at 10,15,25 mg/ml concentrations. Maximum inhibition (13-14 mm) was found in 12,13 and 15 against *S. albus*. Compounds 4,7,8,9 showed moderate activity against *S. aureus*. No significant activity was displayed by other compounds.



Scheme 3.

Anti-fungal activity

The same compounds were tested for their anti-fungal activity against *candida albicans*, *aspergillus niger* and *alternaria alternate* at concentration of 30 mg/ml using sabouraud dextrose agar media. Compounds 12,13 and 15 were found to be moderately active against *candida albicans* and *aspergillus niger*. All the other compounds did not show significant activity against the fungi at the concentration used.

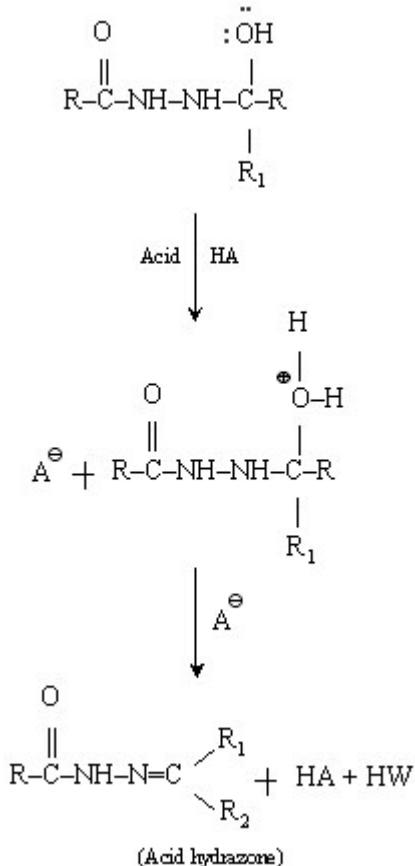


Chart 1: [Mechanism of formation of new acid hydrazones]

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