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Carbene Based Palladium-catalyzed Mizoroki-Heck Reaction

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ABSTRACT

Different perhydrobenzimidazolinium chloride salts were used as a ligand for PdcatalysedMizoroki-Heck reaction. The *in situ* generation of carbene is the attractive feature of this catalytic system. The corresponding Mizoroki-Heck products were obtained in good yield.

Key words: Carbene, Mizoroki-Heck reaction, Palladium.

INTRODUCTION

C-C bond formation is one of the most fundamental and important reactions in the area of organic synthesis and Pd catalysis is the back bone of this bond formation¹. Palladium complex with phosphine-assisted approach² has become a classical and well-established method to give excellent results in most cases. Yet significant drawbacks remain for this approach like phosphine ligands are expensive, toxic, and unrecoverable while handling. In large-scale industrial production, the phosphines might be a more serious economical burden than even palladium itself. Another reason is the poor reactivity often associated with fully ligated complexes of palladium³. Further 2 limitations are found in that amine or carbonate bases and often other additives, usually silver or thallium salts, must be screened in order to optimize the yield⁴.

Recently, N-heterocyclic carbenes (NHCs) came to light as efficient ancillary ligands than phosphine. Increased stability of the active catalyst, lower operational catalyst loadings, tolerance of increased reaction temperatures etc5-7 are the main advantages of carbene based Pd catalysis. Although the nature of the NHC ligand on complexes has a tremendous influence on the rate reactions. of catalyzed the use of perhydrobenzimidazolinium ligand in Heck and Suzuki reactions have been already reported. Perhydrobenzimidazolinium chlorides (3 a, b, c), containing a saturated benzimidazole ring offers in situ palladium-carbene based catalytic system for Heck and Suzuki cross-coupling with improved yields.

Other important Pd-catalyzed C-C coupling reactions are the Mizoroki-Heck [8] and Sonogashira⁹⁻¹². In order to extend the use of this

catalytic system we have used this system for Mizoroki-Heck arylation of alkenes.

EXPERIMENTAL

All the chemicals were purchased from Sigma Aldrich, Acros or Fluka and all the experiments were carried out under nitrogen unless noted. NMR spectra were recorded on standard Bruker 300WB spectrometer with an Avance console at 300 and 75 MHz for ¹H and ¹³C NMR respectively. The residue was purified by flash chromatography with hexane/ethyl acetate.

A high pressure tube was charged with the aryl halide (1 mmol), sryrene (1.5 mmol), salt (3a-c, 2% molar ratio), base (2 mmol), Pd (OAc)₂ (see Table) and DMF/water (3:3 mL). The mixture was stirred at 120 °C and the reaction progress was monitored by GLC. The reaction mixture was cooled to room temperature and an aqueous solution of 1M HCl (10 mL) was added. The mixture was poured into ethyl acetate (10 mL) and washed with 2M HCl (2 x 10 mL) and H₂O (2 x 30 mL). The organic layer was dried over MgSO₄, and the solvents were removed under vacuum. The residue was purified by flash chromatography with hexane/ethyl acetate.

RESULTS AND DISCUSSION

The catalytic activity of the palladiumcarbene based catalytic system was optimized for the Mizoroki-Heck cross-coupling reaction of 4bromoacetophenone 1a with *tert*-butyl acrylate 1b (Figure 1 and Table 1). The reaction was carried out in dimethylformamide/water solvent system using different bases (e.g. NEt_3 , iPr_2NH , NaOAc and K_2CO_3).

According to table 1, Pd/carbine system worked well in the DMF/H₂O solvent system with K_2CO_3 for Mizoroki-Heck cross-coupling reaction between 4-bromoacetophenone 1a with *tert*-butyl acrylate 1b and offersthe respected product in high yield in case of 3a. We took 3a as an ideal salt for further study of Mizoroki-Heck cross-coupling reaction where we took different aromatic halides and styrenes. In most of the cases the corresponding Mizoroki-Heck cross-coupling products were obtained in good to high yield.

Hydroxylatedstilbenoids, specially resveratrol [3,4',5-trihydroxy-(*E*)-stilbene], are a very important class of natural products with important biological activity¹³. The stilbenoid unit is used in molecular photonics and optoelectronics¹⁴. In addition, methoxylatedstilbenoids have shown activity against several human cancer cell lines¹⁵ and are potent CYP1B1 inhibitors valuable for the development for chemopreventive and therapeutic agents for cancer¹⁶. We have also synthesized 3,4',5trihydroxy-(*E*)-stilbene¹⁷ (Table 1, Entry 9) using Pd/ carbine catalytic system by arylation of 4methoxystyrene with 3,5-dimethoxyiodobenzenein in high yield.

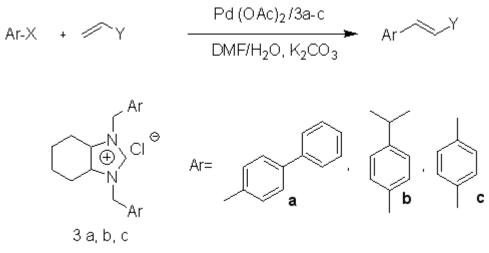


Fig. 1: Mizoroki-Heck cross-coupling reaction

Entry	Aryl halide	Styrene	Salt	Yield (%) ^{a,b}
	(Ar-X)	×γ	(3 a, b, c)	
1.	Br	∕∕CO ₂ tBu	3a	100
	မီ 1a		3b	82
		16	3c	74
2.	O2N Br	∕∕CO2 ^t Bu	3a	92
3.	CI Br	∕~CO2 ^t Bu	3a	98
4.	O2N Br	Ph	3a	87
5.		CN	3a	100
б.		босн₃	3a	79
7.	HO	босн₃	3a	78
8.	M eO	OMe OMe	3a	95
9.	MeO	OMe	3a	98
	ÓMe			

Table 1: Mizoroki-Heck reactions with different aryl halides and styrenes

Reaction conditions:aryl halide (1.0 mmol), styrene (1.5 mmol), base 2mmol, Pd(OAc)2 1% (molar ratio), 3a-c 2% (molar ratio), water /DMF (1:1), 130°C, 2 h. alsolated yields are based on aryl bromide. bthe compound purity was checked by NMR.15

CONCLUSION

We have extended the use of palladiumcarbene based catalytic using perhydrobenzimidazolinium chlorides (Table 1, Entry 3 a-c) as ligand for Mizoroki-Heck crosscoupling reaction with high yield. Synthesis of biologically active compound 3, 4', and 5-trihydroxy-(E) stilbene is a major outcome of our work.

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