



A Novel Synthesis of Biscoumarin Derivatives Catalyzed by $ZnCl_2$ under Solvent-Free Conditions

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ABSTRACT

Zinc chloride ($ZnCl_2$) is used as an efficient catalyst condensation of 4-hydroxycoumarin and with various aromatic aldehydes leading to the formation of biscoumarin derivatives. Easy isolation of the products, short reaction times, excellent yields and reusability of catalyst was the advantage of this method.

Key words: Bis-coumarins , $ZnCl_2$, Aromatic Aldehydes, Solvent-free Conditions.

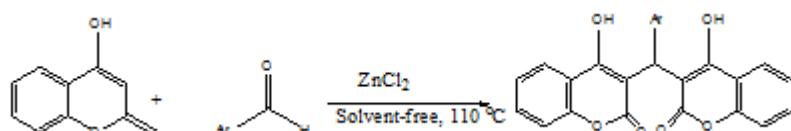
INTRODUCTION

Biscoumarin and its derivatives are of considerable interest due to their biological activities such as urease inhibitors¹, HIV inhibitory activity², anticoagulant activity³, treatment of thrombosis⁴.

From among various catalyst for synthesis of Biscoumarins, the following catalysts can be referred such as sulfuric acid, phosphorus pentoxide, aluminumchloride, iodine, and

trifluoroacetic acid are employed⁵⁻⁶, molecular iodine⁷, $[bmim][BF_4]$ ⁸, tetrabutylammonium bromide (TBAB)⁹, $Zn(\text{Proline})_2$ ¹⁰, sodium dodecyl sulfate (SDS)¹¹. Previously, we have synthesized a number of heterocyclic compounds¹²⁻²².

Herein, we report an efficient method for the synthesis of some biscoumarin derivatives from aromatic aldehydes and 4-hydroxycoumarin catalyzed by Zinc chloride under solvent-free conditions (Scheme 1).



EXPERIMENTAL

All chemicals were obtained from Merck or Flukawithout further purification. Silica gel SILG/UV 254 plates were used for TLC. IR spectra were measured on a Shimadzu IR-470 Spectrophotometer. ¹H NMR spectra were determined on Bruker400 DRX AVANCE instrument at 400 MHz, respectively.

General procedure for preparation of H1-H4

A mixture of aldehyde (1 mmol), 4-hydroxycoumarin (1 mmol), ammonium acetate (3 mmol) and ZnCl₂(15mol %) as a catalyst was stirred at 110 °C for 35 min. The progress of reaction was monitored by TLC. After finishing, recrystallized from ethanol 95% to give pure products (H1-H4)

3,3'-[**(3-Nitrophenyl)methylene]bis(4-hydroxy-2H-chromen-2-one) (H1)**

White crystals, Yield: (89%), mp 233-235°C. IR (ν_{max} /cm⁻¹) (VBr): 3120(arom. CH Str.); 1680,1630 (C=O Str.); 1540 , 1340 (NO₂ Str.) ; 1490(C=C Str.).¹H-NMR (400.13 MHz CDCl₃): δ=6.15(H, s, CH); 7.40-7.61(6H, m, 6CH); 7.69(2H, t,³J=7.6 Hz, 2CH); 8.03(H, d,³J=7.6 Hz, CH); 8.09(H, s, CH); 8.12(H, t,³J=7.6 Hz, CH); 8.17(H, d,³J=8.0 Hz, CH); 11.40(H, s, OH); 11.60(H, s, OH).

3,3'-[**(4-Nitrophenyl)methylene]bis(4-hydroxy-2H-chromen-2-one) (H2)**

White crystals, Yield: (87%), mp230-232°C. IR (ν_{max} /cm⁻¹) (VBr): 3140(arom. CH Str.); 1670,1610 (C=O Str.); 1580 , 1370 (NO₂ Str.) ; 1480(C=C Str.).¹H-NMR (400.13 MHz CDCl₃): δ=6.13(H, s, CH); 7.41-7.47(6H, m, 6CH); 7.69(2H, t,³J=8.0 Hz, 2CH); 8.02(H, d,³J=7.6 Hz, CH); 8.10(H, d,³J=7.8 Hz CH); 8.20(2H, d,³J=8.8 Hz, 2CH); 11.39(H, s, OH); 11.59(H, s, OH).

3,3'-[**(2-Chlorophenyl)methylene]bis(4-hydroxy-2H-chromen-2-one) (H3)**

White crystals, Yield: (90%), mp225-223 °C. IR (ν_{max} /cm⁻¹) (VBr): 3120(arom. CH Str.); 1660, 1620 (C=O Str.); 1490(C=C Str.). ¹H-NMR (400.13 MHz CDCl₃): δ=6.16(H, s, CH); 7.24-7.39(5H, m, 5CH); 7.41(2H, d,³J=8.0 Hz, 2CH); 7.49(1H, d,³J=8.0 Hz, CH); 7.64(2H, t, ³J=7.6 Hz, 2CH); 8.04(2H, m,2CH); 10.95(H, s, OH); 11.67(H, s, OH).

3,3'-[**(2-Fluorophenyl)methylene]bis(4-hydroxy-2H-chromen-2-one) (H4)**

White crystals, Yield: (85%), mp 215-217°C. IR (ν_{max} /cm⁻¹) (VBr): 3130(arom. CH Str.); 1675, 1605 (C=O Str.); 1510(C=C Str.). ¹H-NMR (400.13 MHz CDCl₃): δ=6.13(H, s, CH); 7.18-7.37(8H, m, 8CH); 7.43(1H, d,³J=8.2 Hz, CH); 7.65(2H, d,³J=8.2 Hz, 2CH); 8.03(2H, dd, ³J=28.0 Hz, ³J=8.0 Hz, 2CH); 11.33(H, s, OH); 11.56(H, s, OH).

RESULTS AND DISCUSSION

We have been able to introduce an efficient method for the synthesis of biscoumarin derivatives via condensation of 4-hydroxycoumarin with various aromatic aldehydes. Therefore, reported ZnCl₂as catalyst which could provide an efficient, cheap, easy separationunder solvent-free condition for the synthesis of biscoumarins with high yield.

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