

ORIENTAL JOURNAL OF CHEMISTRY

An International Open Free Access, Peer Reviewed Research Journal

www.orientjchem.org

ISSN: 0970-020 X CODEN: OJCHEG 2014, Vol. 30, No. (1): Pg. 361-363

Synthesis and Characterization of 3-(1-hydroxy naphthalene-2-yl)-5-(furan-2-yl)-1-substituted pyrazolines

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http://dx.doi.org/10.13005/ojc/300150

(Received: December 20, 2013; Accepted: January 10, 2014)

ABSTRACT

2-acetyl-1-naphthol 2 is prepared by Modified Nenchi's method which on treatment with furfuraldehyde and KOH gives 1-(1-hydroxy naphthalen-2-yl)-3-(furan-2-yl) prop-2-ene-1-ones 3 in excellent yield. The chalcone 3 when subjected to hydrazine / phenyl hydrazine/ semicarbazide / 2,4 dinitro phenyl hydrazine / isonicotinic acid hydrazide in DMF solvent gives 3-(1-hydroxy naphthalene-2-yl)-5-(furan-2-yl)-1-substituted pyrazolines 4, 5, 6, 7 and 8 in 35-45% yield. The structural assignments to the compounds 4, 5, 6, 7 and 8 are based on their elemental analysis and spectral data.

Key words: Synthesis, Characterization, Pyrazolines.

INTRODUCTION

The literature survey reveals that pyrazoline derivatives have been studied extensively because of their ready accessibility, diverse chemical reactivity, broad spectrum of biological activity¹⁻⁹ and variety of Industrial applications¹⁰⁻¹². Pyrazolines with sulphonamidoaryl substituent at 3-position show cerebroprotective¹³, antidepressant activity¹⁴, antiimplantation activity¹⁵, hypoglycemic activity¹⁶. Due to this vital biological roll of pyrazoline derivatives, it was thought of interest to synthesize titled pyrazoline derivatives of the titled compounds. It has been observed that substituted flavanones are the best starting compounds for the preparation of 4-aroyl derivatives of pyrazoline. Present work deals with the synthesis of some new pyrazolines and their characterization by spectral analysis (IR, ¹H NMR)

EXPERIMENTAL

All the melting points were taken in silicon oil bath with open capillary tubes and are uncorrected. ¹ H NMR spectra were recorded on a Brucker AC300 FNMR spectrometer(300MHz), using TMS as an internal standard. IR spectra were recorded on a Nicolet-Impact 400 FT-IR spectrometer. Thin Layer Chromatography on silica gel-G, was used to check the purity of the compounds. Microanalysis of nitrogen was obtained on Colman 29-N analyzer.

Experimental Section Procedure for the synthesis of 2-acetyl-1naphthol 2

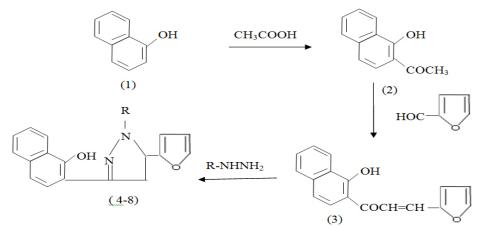
In hot glacial acetic acid (80ml), fused $ZnCl_2(50 \text{ gm})$ was added and refluxed till dissolved, then powdered 1-naphthol (30gm)was added and

the mixture was refluxed for about 8 hours then cooled & poured in acidulated water. The solid obtained was filtered, washed, dried and recrystallized from rectified spirit to obtain compound 2. Physical data of the compound is given in table 1.

2: IR (KBr) : 1650 (C=O), 3412 (-OH)

NMR (CDCl₃ + DMSO-d6) : δ 2.35 (S,3H,CH₃), δ 7.11-6.88 (m, 6H, Ar-H), [^] 9.83 (S,1H,-OH)

S.	Comp ound	Molecular	R	Melting	%	% Nitrogen		R.F.
No.	No	formula		Point °C	Yield	Found	Calculated	Value
1	2	C ₁₂ H ₁₀ O ₂	-	98ºC	72%	-	-	
2	3	C ₁₇ H ₁₂ O ₃	-	126ºC	71%	-	-	
3	4	C ₁₇ H ₁₄ N ₂ O ₂	Н	110ºC	38%	10.00	10.07	0.79
4	5	C ₂₃ H ₁₈ N ₂ O ₂	C ₆ H ₅	100ºC	40%	7.81	7.91	0.88
5	6	C ₁₈ H ₁₅ N ₃ O ₃	CONH,	140ºC	45%	12.99	13.08	0.87
6	7	C ₂₃ H ₁₆ N ₄ O ₆	C ₆ H ₃ N ₂ O ₄	120ºC	40%	12.56	12.61	0.78
7	8	$C_{23}H_{17}N_{3}O_{3}$	C₅H₄NCO	117ºC	35%	10.91	10.97	0.84



 $R = H, C_6H_5, CONH_2, C_6H_3N_2O_4, C_5H_4NCO$

Synthesis of 1-(1-hydroxy naphthalen-2-yl)-3-(furan-2-yl) prop-2-ene-1-one (3)

2-acetyl-1-naphthol (0.01mole) and furfuraldehyde (0.02 mole) were added in ethanol solvent (20ml). To this mixture KOH (10%, 10ml) solution was added dropwise with constant stirring. The reaction mixture was kept overnight. Then the mixture was poured over crushed ice & little HCI. The product was filtered and recrystallized from ethanol to obtain the compounds (3). The physical data is given in Table 1

3: IR (KBr):1650 (C=O), 3412(-OH), 1155(C-O-C) NMR (CDCl₃ + DMSO-d6): δ 6.98-7.46 (m,9H, Ar-H), δ 8.057 (d,1H =CH), δ 8.099(d,1H =CH) δ 9.63 (S,1H,-OH)

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Synthesis of 3 - (1-hydroxy naphthalen - 2 - yl)-5-(furan-2-yl) - 1-substituted pyrazolines (4-8)

1-(1-hydroxy naphthalen-2-yl)-3-(furan-2-yl) prop-2-ene-1-ones (0.01mole) & hydrazine / phenyl hydrazine/ semicarbazide / 2,4 dinitro phenyl hydrazine / isonicotinic acid hydrazide (0.01 mole) were added to DMF (20 ml) and refluxed for 2 hours. The cooled reaction mixture was diluted with water & the semisolid so obtained was triturated with ethanol to get a solid which was recrystallised from ethanol-acetic acid mixture to get titled pyrazolines in 45-51% yield . The physical data is given in Table 1

4: IR (KBr):3000 (N-H), 3402(-OH), 1151(C-O-C) NMR (CDCl₃ + DMSO-d6) : δ 6.90-7.80 (m,6H,

Ar-H), δ 8.051 (d,1H =CH), δ 8.091(d,1H =CH) δ 9.62 (S,1H,-OH)

5: IR (KBr):1600 (C=C), 3410(-OH), 1148(C-O-C) NMR (CDCl₃ + DMSO-d6) : δ 6.89-7.90 (m,11H, Ar-H), δ 8.049 (d,1H =CH), δ 8.081(d,1H =CH) δ 9.53 (S,1H,-OH)

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 $\begin{array}{l} \textbf{6: IR} \ (KBr):1723 \ (C=O), \ 3405(\text{-OH}), \\ 1160(\text{C-O-C}) \ \text{NMR} \ (\text{CDCI}_3 + \text{DMSO-d6}): \delta \ 6.81\text{-}7.40 \\ (m, 6H, \ Ar\text{-}H), \ \delta \ 8.041 \ (d, 1H = \text{CH}), \ \delta \ 8.079(d, 1H = \text{CH}) \\ \text{=CH} \ \delta \ 9.49 \ (S, 1H, \text{-OH}), \ 6.6 \ (S, 2H, \text{NH}_2) \end{array}$

 $\label{eq:constraint} \begin{array}{l} \textbf{7:IR(KBr):}1530,1550(NO_2),3412(OH),\\ 1155(C\text{-O-C})\ NMR\ (CDCI_3+DMSO\text{-}d6): \delta\ 6.81\text{-}7.60\\ (m,9H,\ Ar\text{-}H),\ \delta\ 8.045\ (d,1H=CH)\ ,\ \delta\ 8.081(d,1H=CH)\ \delta\ 9.43\ (S,1H,\text{-}OH) \end{array}$

8: IR (KBr):1611 (C=O), 3477(-OH), 1170(C-O-C) NMR(CDCl₃ + DMSO-d6): δ 6.95-7.81 (m,10H, Ar-H), δ 8.043 (d,1H =CH), δ 8.083(d,1H =CH) δ 9.50 (S,1H,-OH)

ACKNOWLEDGEMENTS

The author is thankful to the Director, Govt. Vidarbha Institute of Science and Humanities, Amravati.

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