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Determination of Aniline Blue dye by Flow Injection Analysis With Home Made Valve

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ABSTRACT

Flow injection unit with merging zone technique was designed for aniline blue determination spectrophotometrically. This unit consists of a valve was designed for dye injection. The scope of the study is to find the optimum conditions :volume of sample, flow rate of carrier, dead volume, dispersion coefficient and reproducibility. They found 117µl, 4.3 ml/min, zero dead volume, 2.62 and high reproducibility for six times respectively. The range of aniline blue concentration by flow injection analysis was 0.1 – 8 ppm and detection limit 0.1 with R^2 = 0.9952 and 0.01 – 16 ppm for spectrophotometric method with R^2 =0.9947 with detection limit 0.01.

Key Words : Merging zone, Valve , Aniline blue, Dead volume, Dispersion, Reproducibility.

INTRODUCTION

Flow injection analysis (FIA) is one of recent techniques are established by Ruzickaan Hansen in 1975 at Denmark and Stewart in USA that is the first generation ¹. The second generation of FIA is sequential injection analysis (SIA)² and Bead Injection (BI) is the third generation of FIA techniques³. The main skeletal of FIA system is: injector valve, peristaltic pump, reaction coil, tubing manifold and detector ⁴. FIA technique has ability to analyze very low volumes of samples , high sampling output and low detection limit ⁵, also has very good conditions of dead volume and dispersion⁶. FIA depends on sample injection in same volume into the carrier stream, controlling on dispersion zone to form product⁷. The sample injected through the valve by carrier solution move continuously to reaction coil for mixing then a product formed moving to the detector and response recorded as physical parameter changing⁸. The result signal recorded as single peak which has height, area, time of sample analyzing appearance and the width of peak ⁷. FIA can be modified by many ways: revers FIA⁹, merging zone¹⁰ and stop flow ¹¹ and can be connected with chemluminescence¹², fluorescence¹³ and luminescence¹⁴.

Flow injection valve is an instrument used for injection the very small equal volumes of samples by the carrier rapidly⁶. The valves work electrically according to electronic program are the most development and its superior in small injected volume reproducibility and high speed injection³ and valves in high performance liquid chromatography (HPLC) are suitable for flow injection analysis ^{8.} There are many valves made in Iraq, one the valve extend for sample and reagent which used for determination trace concentration of acids^{15,16}. Different design of valve extends for the sample and reagent which it used in FIA system for determining phenol¹⁷ and adsorbed on rice husk ¹⁸. A new design valve extends sample and three reagents injection using for determination vanadium ¹⁹. A new valve with high performance two levels valve was made to extend for eight material injection²⁰.

Dyes are organic compounds that absorb light in specific areas of the visible spectrum²¹, Aniline blue dye is used for dying wool and cotton directly and used in dye industries that make a contribution to environmental pollution²². Aniline Blue dye is very soluble in water and itis an acidic dye belongs to triphenyl methane class²³ which are dyes have fused benzene rings (chromophore) and the auxochromes are - NH_2 , NR_2 and $-OH^{24}$.

In this research aniline blue dye was determined by flow injection techniques using homemade valve and study the effective conditions on this determination then compared with convenient spectrophotometry method and This leads to the possibility of using flow injection analysis technique for the study of dyes in all their interactions, for example adsorption of dye^{25,26,27}.

EXPERIMENTAL

- 1. Chemicals: aniline blue (BDH)(50 ppm). Distilled water as carrier solution .
- Equipments : prestatic pump (ISM796, Switzerland), Homemade injection valve, flow injection cell (Helma), spectrophotometer (Aple PD-303UV, Japan), Spectrometer (T80 UV/VIS China), Kompensograph (C1032, simens, Netherlands) and Teflon tubes.
- FIA unit and Valve Work : FIA unit was designed for determination of aniline blue dye. This designed and installed were in FIA lab, chemistry dept., college of science, university

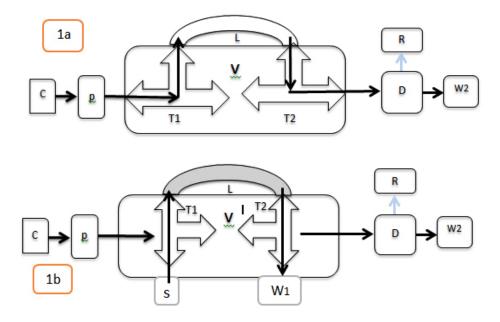


Fig. 1: (a) FIA and valve design for aniline blue determination at carrier loading (b) at sample loading (C- water carrier P- pump V- valve (T1,T2 : three way, L: sample loop) S- injected sample W1- waste of sample W2- waste after mixing D- detector R- recorder).

of Babylon, Iraq. The carrier distilled water flow through the pump to the valve and fill all parts of the unit outing to W2 , this case gave the base line as showed in Figure 1a. The second step is the injection of the sample from T1 to loop sample L and the waste outing from W1 as showed in Figure 1b. T1 and T2 moving to first case in figure 1a and the carrier pump the sample and the absorbance was measured at 580 nm . Recorder reads this absorbance as a peaks .

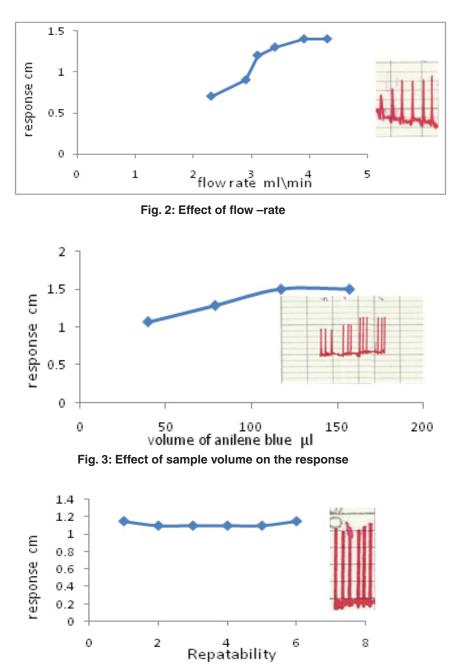


Fig. 4: the reproducibility for 6 ppm of aniline blue sample at flow rate 4.3 ml/min , sample volume $117\mu l$, 580 nm

RESULTS and DISCUSSIONS

Effect of Flow Rate

Flow rate is one of the important factors affecting in FIA which make by change the carrier flow via pump²⁸. Ruzicka and Hansen show that the occurrence of dispersion be less at low flow rate because the retention time of sample increased before moving to the detector lead to increasing interaction between sample and reagent³. Also, peaks at low flow rate are a broad, double and distorted²⁹. Sometime, high flow rate reduce the response³⁰. At 580 nm and 117 microliter , 6 ppm from aniline blue was injected in FIA unit each time

in different flow rate (2.3, 2.9, 3.1, 3.4, 3.9) and 4.3 ml per min) and the response was recorded as showed in Figure 2.

When the flow rate increased ,the response increase. In low flow rate the carrier made a dilution for aniline blue dye more than in high flow rate 31 . In high flow rate 3.9 and 4.3 ml/min , the flow rate was not affected the response 32 . 4.3 ml/min was the best value with the highest response 1.4 cm.

Effect of sample volume (sample loop)

Sample volume is important influences affecting on dispersion process which is study by

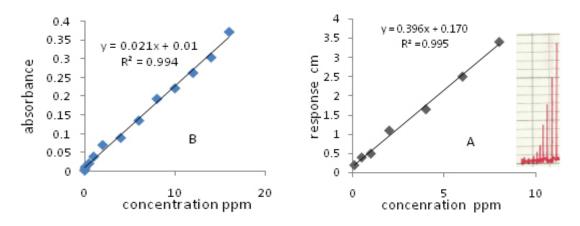


Fig. 5: The calibration curve of aniline blue dye by A: FIA method , B : by spectrophotometry method.

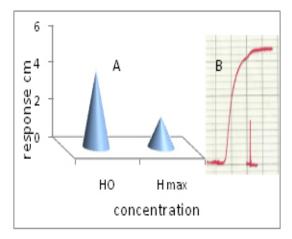


Fig. 6: Effect of dispersion parameter, a: response according with and without dispersion, b: the recorded response

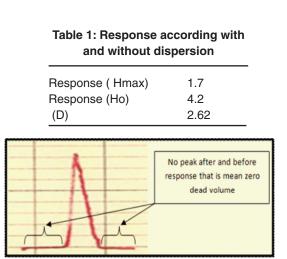


Fig. 7: response without dead volume

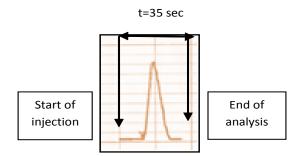


Fig. 8: the time for one sample analysis

using different length loops³⁰. Increasing sample volume lead to increase response (peak height)33) but the access of sample volume make incomplete dispersion that caused broad and Double peaks³⁴. At 580 nm, 4.3 ml/min flow rate and 6 ppm of aniline blue dye, the effect of sample volume (changing loop) was studied and response was recorded for each loop. the volume of sample under study were 3.9.25, 78.5, 117 and 157 microliter. The result showed in figure 3. When the sample volume Increase the response was increased, after 117 and 157 µl the response remain constant. This effect is similar to results for FIA unit used for determination of water toxicity ³⁵ and determination of magnesium ³⁶. Although the response for 117 and 157 are equal, but 117 µl was dependent as the best value because the volume as small as will be more economical .

Repeatability

Repeatability is a successive injections for the same concentration of sample under the optimum conditions to study efficiency of FIA unit³³. At 580 nm , 4.3 ml/min flow rate ,117 microliter and 6 ppm from aniline blue , the reproducibility was studied . Its mean the same readings can be gain it in case of inject one sample several times (6 times). Reproducibility can be seen in figure 4 it shows the repeating run with response. The unit gave a good reproducibility which agree with results for determination of adsorbed phenol on wheat husk ³⁷ , proscopisfarcta berry³⁸ , for alkenes determination by merging zone technique ³⁹ and alkenes determination by sequential technique⁴⁰.

Calibration

A series concentration of aniline blue was taken to make calibration curve . The calibration was

studied at 580 nm , flow rate 4.3 ml/min and sample volume 117 μ l Detection limit was 0.1 ppm , range was (0.1-8) ppm and R² was 0.9952. Compare with spectrophotometry method .Detection limit , range and R² are (0.01-16) ppm , 0.01 ppm and 0.9947 respectively as in figure 5.

Dispersion coefficient (D)

It is the ratio of the concentration of the sample before and after the dispersion process takes place. Dispersion coefficient is defined by the equation $(D = H^{\circ} / H_{max})$

Where H° = peak concentration in injection volume without dispersion , H_{max} peak concentration at detector with dispersion³⁸. D is influenced by three interrelated and controllable variables: sample volume, tube length, and pumping rate³⁹. The dispersion coefficient was studied at 495 nm , flow rate 3.4 ml/min and sample volume 275µl the absorbance was measured with dispersion H_{max} and without dispersion H^o as showed in table 1.The absorbance experimentally decrease with dispersion . The dispersion coefficient (D) was calculated from following equation ⁴¹ as showed in figure 6.

Dead volume

It is the volume that remained in valve at loading sample or reagents and it leads to appear faulty peak :double, distorted or signal in base line, In this case the valve is imperfect but if the dead volume equal zero, the valve is perfect³⁰. The homemade valve which used in this work has a zero dead volume. Where all responses start from zero baseline for one response. This case is perfect in our work and this accomplished to the past work in FIA lab.: methylene blue determinationby one loop valve FIA system⁴², three loop valve for determination Sn⁴³ and Mn⁴⁴ and four loop valve for Vdetermination⁴⁵, all these valves have zero dead volume and figure 7 shows the idea of dead volume with response.

Sampling Output

Sampling output was computed in hour by measuring time required for one run. The sample output is 102 samples per hours as showed in figure 8.

CONCLUSIONS

FIA unit which used in this work was Cheap and gave a good results for determining aniline blue by using a small volume of the sample and very good output sampling comparing with spectrophotometric method. ACKNOWLEDGEMENT

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