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# Evaluation of the Role of Cetyltrimethylammoniumbromide (CTAB) and Acetylenicglycol (AG) Admixture on Fly Ash Based Geopolymer

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## **ABSTRACT**

Alkali activation of fly ash with proper mix design and correct formulations can exhibit very good strength and chemical resistance and other potentially valuable characteristics similar or even better than conventional Portland cement. Utilization of fly ash activated system as binder material in construction industry can limit the consumption of virgin materials required for the production of cement as well less energy intensive process in the manufacturing stage leads to lower greenhouse gas emission. Goeopolymeric products with characteristic properties can be achieved on a laboratory scale level by optimizing the process conditions using selected source material with the addition of admixture. On contrary, Usage of variable source material in a large scale preparation, issues like flow ability, control of setting time are always present that need to be studied. This study was conducted on leaching of fly ash by hybrid activator solution namely sodium silicate, sodium hydroxide, for preparing geopolymer matrix wherein geopolymer (GPC) mortar was made with the addition of organic admixture Cetyltrimethylammonium bromide (CTAB), Acetylenic glycol (AG) as flow ability enhancer with dosage ranging from 0-2% by keeping the Liquid/Solid as 0.45.The Compressive strength of the mortar in the order 30 Mpa was achieved with good flow property by the addition acetylenic glycol and the zeta potential value was found to be similar in the order of control specimen(-15.4mv). Analysis of SEM microstructure of the matrix revealed that binding gel of type Na-S-H are found to be more on AEG added matrix than that of CTAB.

Keywords: fly ash, Geopolymer, compressive strength, slump, SEM/EDAX, Zeta potential.

#### INTRODUCTION

Geopolymer has recently emerged as a greener binder material for sustainable construction <sup>1</sup>. It is well known for its excellent properties such as fire and acid resistance, high compressive strength,

low shrinkage and solidification of heavy metal wastes<sup>2-5</sup>. Geopolymers are used in the toxic metal encapsulation and also used for the preparation of zeolite by incorporation of sodium hydroxide in the reactive system.<sup>6</sup> It has tremendous potential applications of new areas in recent years<sup>7,8</sup>. One of

the best ways is to synthesize, geopolymer a binder made by alkali-activated solution because of its low-cost, environment-friendly, energy-saving and resource-recycling benefits.

The alkali activation of aluminosilicate materials is a chemical process that transforms partially or wholly vitreous structures into compact cementitious skeletons. The key processes involved in the synthesis of geopolymers are: (a) Dissolution of fly ash or similar alumino-silicate precursors to provide the Aluminium and all or part of the Silicon constituents needed. (b) hydrolysis reaction to generate aluminum and silicate species and finally (c) condensation of these species and/or silicates from the activator to build up geopolymeric network structures. Geopolymers (GP) with frameworks of [SiO<sub>4</sub>]<sup>4</sup>"and [AlO<sub>4</sub>]<sup>5</sup>"tetrahedral linked alternately by sharing all the oxygen atoms, are generally amorphous aluminosilicates or semi-crystalline zeolites depending on the alkaline activator used in their manufactures and reaction conditions. The negative charge of [AlO<sub>4</sub>]<sup>5"</sup>tetrahedron on skeleton is balanced by extra-framework cations of Na+, K+ and Ca<sup>2+</sup>ions. <sup>9</sup> Corresponding to different Si/Al ratios, the geopolymers are composed of network structures made up of polysialate (-O -Si- O- Al -O), polysialate siloxo (-O- Si- O -Al -O- Si -O-), and polysialate disiloxo (- O -Si- O- Al- O -Si- O -Si- O).

The interaction of various process parameters, nature of admixtures on the compressive strength and workability of Geopolymer products are complex and hence final Geopolymer products exhibit different performance properties viz. high compressive strength, high thermal resistance, low shrinkage & low thermal conductivity, high level of resistance to a range of different acids and salt solutions etc. It is also important to note that all Geopolymer products will result all of these properties or in other words, no single formulation matrix posses all the properties. Hence formulations

recipes are required to tailor made Geopolymer mixes to attain the required specifications and technical performance. This will be possible only with thorough understanding of the raw material reactivity, chemistry, role of admixtures and reaction conditions.

The commercial superplasticizers viz., lignosulphonate, sulphonate naphthalene formaldehyde (SNF), sulphonate melamine formaldehyde (SMF), and polycarboxylates which are used in the production of concrete that cement particles by electrostatic repulsion through adsorption process, enhance workability, giving high performance and improve durability. The mechanism and working nature of superplasticizers in OPC have been well reported in literature. <sup>10-14</sup>. Such type of chemical admixtures does not work in Geopolymer system because of high alkaline conditions. <sup>15-20</sup>

Flyash activated with 85% of 12.5 M NaOH solution and 15% of waterglass with commercial chemical admixtures such as lignosulphonates, melamines(first and second generation products) and polycarboxylates (latestgeneration) were reported. The flow table spread results showedthat the polycarboxylates admixture gave the highest spread, AAFA without admixture came in the second place, melamines admixturecame in the third place, whilst lignosulphates admixturecame in the last place. Studies using commercial superplasticizers such as Sika Viscocrete-3430 (CONPLAST SP-430), showed that these high range water reducers had negative on the compressive strength of Fly ash based geopolymer but improved workability of the concrete. 21-22

The effect of third generation SP such as vinyl copolymers, poly acrylate copolymers along with modified form of melamine, naphthalene, Polycarboxylate ester based, on alkali activated slag and fly ash based geopolymers have been

Table.1: Chemical composition of low lime fly ash (Indian)

Chemical	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	CaO	Fe <sub>2</sub> O <sub>3</sub>	K <sub>2</sub> O	MgO	Na <sub>2</sub> O
Component (wt %) of fly	47.55	33.45	2.099	10.17	1.65	0.005	0.015

reported.<sup>23</sup> It was found that addition of N-based SP improved the workability upto 2% mass of binder without affecting the strength. Studies on evaluation of PC based SP on fly ash Geopolymer showed appreciable reduction in the strength around 20 to 35% with reference to original concrete. Similar such conditions are existing for geopolymer from slag and fly ash base combination. Reports on the addition of fly ash with slag improving workability of alkali activated slag system. <sup>24,25</sup> On the contrary report on Indian fly ash did not improve the workability even at 40% fly and 60% slag base.

Till date suitable admixture not deleterious for compressive strength, working under high alkaline conditions has not been reported in literature and further research in this area is necessary in order to scale up the Geopolymer products.

The present study aims to investigate the suitability of organic compounds CTAB and AG working under NaOH and sodium silicate combined activated fly ash Geopolymer system.

## **MATERIALS AND METHODS**

## Physico-chemical properties of fly ash

The class F fly ash (Indian) are characterized with regard to their mineralogy, microstructure and morphology. The chemical composition of the low

Fig.1(a): Molecular structure of CTAB

Fig.1(b): Molecular structure of AG

lime fly ash is determined by X-ray Fluorescence (XRF) analysis and is summarized in Table1. XRD (Siemens D 501 diffractometer using Cu K\_radiation) diffractogram. The microstructure of the source material (Fly ash) and Geopolymer mix were taken using Scanning Electron Microscope.

The chemical composition of fly ash(Indian) as determined by XRF are according to ASTM C 618. The silica/alumina ratio of the fly ash by mass was (SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub>) 1.42.

## Alkali activator

A mixture of sodium hydroxide (NaOH) and sodium silicate ( $Na_2SiO_3$ ) solutions was used as the activator solution. The concentration of sodium hydroxide solution was kept constant (12 M) for all mixtures. Sodium silicate solution used in the present study contains mass ratio of  $Si_2O/Al_2O_3$  as 0.5 with chemical compositions 30.7%  $SiO_2$ , 11.4%  $Na_2O$  and 57.9% water. The fine aggregate used was natural sand having a nominal maximum size of 1.18 mm. The activator solutions of desired quantity were prepared a day before and used with other ingredients to enhance reactivity of the solution.

#### Organic admixture

The SP used in this study was cetyltrimethyl ammonium bromide (CTAB) and Acetylenic glycol (AG). CTAB (Chemie Pvt. Ltd., Mumbai), a cationic surfactant having chemical formula  $\rm C_{19}H_{42}BrN$ . This was chosen due to its relative inertness with calcium or sodium ions present in system. The molecular structures of CTAB, AG based SP are shown in Fig.1 (a) & (b). Acetylenic glycol(AG) a nonionic surfactant with a bilateral-symmetrical structure and an acetylene bond at the center of its molecule. Its hydrophobic nature lower the surface tension thereby significant reduction of water sensitivity is expected.  $^{20}$ 

# Geopolymer mix design

Geopolymeric paste and mortar mixtures were proportioned to study the effect of SP dosage. The mixture proportions of paste and mortar are presented in Table 2. The total binder content was constant and constituted one fourth of the total mixture. The mix proportions of the pastes were similar to those of the mortar mixtures with the aggregates excluded.

## Mixing, casting and curing

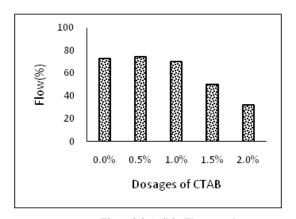
Fine and coarse aggregates prepared in saturated surface dry condition and the binders (fly ash) were dry mixed thoroughly in the mixing pan for 2 minutes. The premixed alkaline activator solution was then added gradually in to the mixer. Mixing was continued for further 3 min. SP's dosage was varied from 0 .5% ,1%, 1.5%,2.0% by mass of fly ash and the whole mix was blended for 4 min. to eliminate the clustering of fine particle.

The geopolymer mortars were prepared to obtain a uniform mortar mixture. Cylindrical moulds, 75 mm in diameter and 150 mm in height, were filled with geopolymer mortar in two layers and compacted on a vibrating table. Cube samples (25x 25x 25 mm³) were casted with geopolymer paste. All the

specimens were sealed with polythene sheets to prevent excess evaporation/loss of moisture. After 24 h of curing , specimens were oven dried for 24 h at 70°C and then demolded.

# **Test methodology**

The workability of fresh geopolymer concrete mixtures were tested by Slump test following ASTM: C1437.Flow of fresh geopolymer mortars were measured in accordance with ASTM C1437-07 to determine workability loss of fresh concretes. A layer of the fresh concrete mix about was poured into the truncated conical mould (top diameter = 100 mm, bottom diameter = 200 mm, height = 300 mm) and tampered 25 times with tamper. The mould was firmly held against its base during the test or foot-rest brazed to the mould. Immediately after filling, the



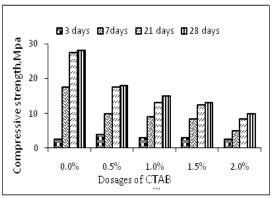


Fig.2 (a) & (b): Flow and compressive strength of geopolymer with different dosages of CTAB

Table 2: Geopolymer mix design

S.	Designation	Mortar mixture quantity(Kg/m³)						Activator solution		
no		FA	Sand	SP	Sodium hydroxide	Sodium silicate	L/S	Na <sub>2</sub> O /Al <sub>2</sub> O <sub>3</sub>	H <sub>2</sub> O /Na <sub>2</sub> O	Si <sub>2</sub> O /Al <sub>2</sub> O <sub>3</sub>
1	AD-O	500	1000	0.0	350	100	0.5	0.43	3.35	0.5
2	AD-C-1	500	1000	2.5	350	100	0.5	0.43	3.35	0.5
3	AD-C-2	500	1000	5	350	100	0.5	0.43	3.35	0.5
4	AD-C-3	500	1000	7.5	350	100	0.5	0.43	3.35	0.5
5	AD-C-4	500	1000	10	350	100	0.5	0.43	3.35	0.5
6	AD-A-5	500	1000	2.5	350	100	0.5	0.43	3.35	0.5
7	AD-A-6	500	1000	5	350	100	0.5	0.43	3.35	0.5
8	AD-A-7	//500	1000	7.5	350	100	0.5	0.43	3.35	0.5
9	AD-A-8	500	1000	10	350	100	0.5	0.43	3.35	0.5

cone was slowly lifted and the unsupported concrete was then slumped. The decrease in the height of the concrete was then measured. The consistency of the mixes was measured using a Vicat needle as described in ASTM C191-08. Values of penetration distances were measured within 5 min. after the preparation of the paste. The test was conducted in room temperature.

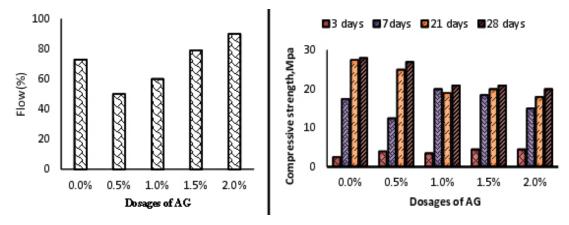


Fig.3 (a) & (b): Flow and compressive strength of geopolymer with different dosages of AG

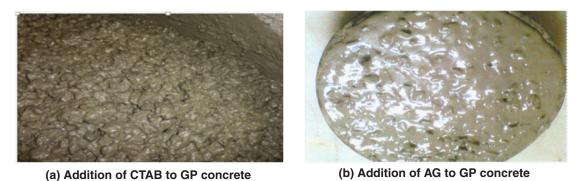


Fig.4: Fresh geopolymer concrete with the addition of (a)CTAB (b) AG

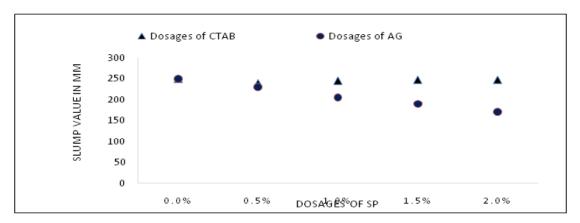


Fig.5: Slump value of geopolymer with varying concentration of CTAB and AG.

Compressive strength test was conducted at different curing regimes 3, 7 and 28 days, with a loading rate of 0.33 MP a/s in a Controls CTM machine. Average of 3 test results was taken for calculations. Malvern Zetasizer (Nano series) was used to measure the zeta potential against the standard potassium tungsto -silicate solution and calculations were made through Zetasizer software. 0.5 weight % of Geopolymer matrices in deionized water was used to measure the zeta potential.

# **RESULTS and DISCUSSIONS**

X-ray diffractogram analysis has shown that the fly ash consists of approximately 70% amorphous materials and remaining mineral phases like quartz, magnetite, anhydrite, anorthite, hematite, and other minerals. Amorphous content in the fly ash are easily dissolve during the initial step of the geopolymerisation process, yielding higher amounts of reactive SiO<sub>2</sub> and Al<sub>2</sub>O<sub>3</sub> to combine during the

agglomerate phase of the geopolymeric reaction. This increases the rate of geopolymerization, thereby giving high compressive strength.

## Flow and strength of GP mixes

The effect of addition of SP on flow of fresh GP mixes of liquid/solid = 0.45 having 12M of NaOH, have been studied and was presented in the Figure (2) & (3). The flow of alkali-activated fly ash concrete with the addition of CTAB decreased with increasing dosage level. The mixes were found to be very stiff. In the case of AG, fresh mix showed slight enhancement in the flow make it more workable even in the reduced water content. The flow and compressive strength data on various curing time with and without addition of SP is shown in the Fig 2(a) & 2(b). and Fig 3(a) &3(b)

It was found that addition of CTAB has not improved the flowability beyond 0.5% and reduction in the compressive strength was found to

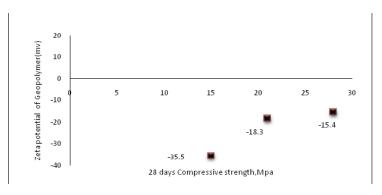
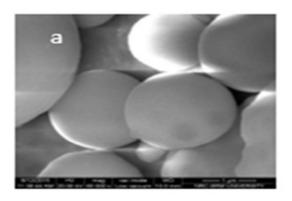


Fig. 6: Zeta potential and compressive strength Geopolymer



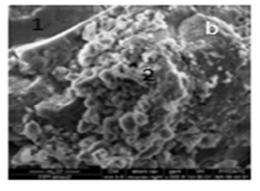
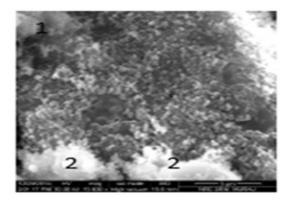
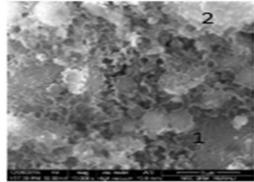


Fig. 7: SEM images of (a)original fly ash and (b) geopolymer without SP

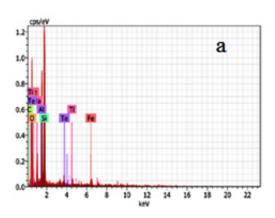




Geopolymer

2. Gel formation over the fly ash

Fig. 8: SEM images of geopolymer with the addition of (a) AG and (b) CTAB



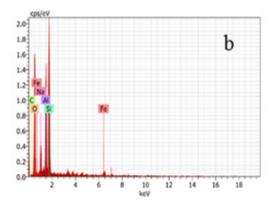


Fig. 9: EDAX for GP mortar with the addition of (a)CTAB (b) AG

be in the order of 35%. In the case of AG-mixes, the compressive strength has not altered much in all the curing regime as well dosage upto 2%.

## Slump test

The fresh fly ash based geopolymer has shown in Fig.4(a) & 4(b). Variations of slump values of the mixes by the addition of SP are shown in the Fig.5. Variation of workability is not so significant in the case of CTAB where as the slump values decreased from 250 mm to 130.mm when AG dosage increased from 0.00 to 2%. Water retarding nature AG may assist the setting property of fresh Geopolymer concrete.

# Zeta potential

The Geopolymer strength characteristics especially Fly ash base depend on the nature of surface charges over the reactive component generally amorphous/glassy content. During the alkali activation, surface layer of the fly ash particles gets deprotonated forming soluble form of  $\mathrm{SiO}_2$  species.

Normally surface layer fly ash particles contain a significant amount of readily leachable elements mostly anionic groups being silicates [-O-SiO<sup>2-</sup>] and aluminates [-O-AIO]. Zeta potential value of fly ashes is in the range of -20 to -25 mv. In our present study the fly ash precursor showed the negative value of -27.8mv.

During the geopolymersiation reaction, activator solution having OH<sup>-</sup> ions react with the aluminate species on the surface of fly ash forming[Al(OH)<sub>4</sub>]<sup>-</sup> and with silicate species form either [SiO(OH)<sub>4</sub>]<sup>-</sup>or [SiO<sub>2</sub>(OH)<sub>3</sub>]<sup>2-</sup>, thereby establishment

of negatively charged double layer. Further Na+ions of the alkaline activator react to form a sodium alumino silicate gel layer, [Na<sub>z</sub>(AlO<sub>2</sub>)<sub>x</sub>(SiO<sub>2</sub>)<sub>y</sub>·nNaOH. mH<sub>2</sub>O], on their surfaces.<sup>26</sup>This reaction step is very important to achieve desirable mechanical strength. While forming Na-Al-Si-H gel the accumulation of more Na<sup>+</sup> ions in the double layer decrease the zeta potential.

In the present study also the same trend was observed showing smaller negative value for zeta potential, (-15.4mv proved that combined activator solution (Sodium hydroxide and Sodium silicate) helped for agglomeration of fly ash particles to facilitate the geopolymersiation reaction. The effect of addition of SP on zeta potential studied under the same conditions of that of the control. The charge distributed over the surface of the SP disturbing the agglomeration of fly ash with activator during the Geopolymeriasation. This was reflected in the zeta potential value measured CTAB admixed Geopolymer paste(-35.5mV) as shown in Fig.6. It may be due to the adsorption of cationic CTAB. In the case of AG, a nonionic compound, shifting the more positive value than that of CTAB but it was lower than that of control. The extent of reaction with the presence of SP is found to be low in all these studies and it was found to be reflected in the compressive strength data.

## **SEM** images

The morphology of fly ash precursors observed from SEM photograph showed that fly ash particles are appeared to be hollow, spherical as thin walled hollow sphere Fig 7(a) & (b). SEM picture of alkali activated fly ash mortar after 28 days of curing time clearly showed that dense microstructure by the formation gel binder. The compacted structure of mortar this study indicated that large amount of reactive product without any crack corresponding to mechanical compressive strength. The addition of SP in the mortar does not affect the microstructure of the Geopolymer formed as shown in Fig 8(a) & (b).

The gel formation over the surface of fly ash particles are clearly seen.

EDAX of specific sample area is shown in Fig.9 Average ratio of Si/Al for fly ash geopolymer was approximately 1.5 was maintained. The data were obtained from the average of five sample areas. The measured Na atomic weights of 12M NaOH mortar of 0.67-0.9.

## CONCLUSION

Short term engineering properties like consistency and flowability of fly ash based geopolymer using organic admixture such as Cetyltrimethylammoniumbromide, acetylenic glycol will help us to find out the time available for placing, compaction and transport of Geopolymer during large scale preparation. The workability of GP concrete was found to be improved than that control while addition AG from 0.5 to 2%. The compressive strength of heat cured (80°C)specimens were not much affected by AG addition whereas around 35% reduction were found in CTAB at all dosage compared to that of control specimens. The results from EDAX indicate that major components in Geopolymer were silicon and aluminum having Si/ Al ratio 1.5 with small amount of sodium.

The shift in value of zeta potential to more positive with and without addition of SP in fly ashwater slurry revealed that addition of SP facilitate the dispersion of fly ash particles. The value of zeta potential observed in the case of geopolymer mortar with addition of SP were found to be more or less same to control specimens which indicates the gel formation has not much affected. Therefore, further collaborative research is required to develop suitable admixture to efficient the geopolymer technology.

# **ACKNOWLEDGMENT**

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